

1 **ECOTOXICOLOGICAL RISK ASSESSMENT AS A DECISION TOOL FOR**
2 **THE VALORISATION OF WINERY WASTEWATER TREATMENT**
3 **BYPRODUCTS TO AGRICULTURAL PURPOSES**

4
5 Silvana Luz^{1,2,3,4}, Mariana Regato⁵, Alexandra Afonso¹, Adriana Catarino^{1,6}, Teresa Santos¹, Patrícia
6 Palma^{1,6}, Javier Rivas², Fátima Carvalho^{1,7}

7
8 ¹Departamento de Ciências e Tecnologias Aplicadas, Escola Superior Agrária de Beja, Instituto
9 Politécnico de Beja, Rua Pedro Soares (IPBeja) - Campus do IPBeja, 7800-295 Beja, Portugal

10 ²Universidad de Extremadura/Depto Ingeniería Química y química física/University Institute
11 IACYS – Avda. de Elvas, 06006 Badajoz

12 ³Centro de Biotecnologia Agrícola e Agro-Alimentar do Alentejo (CEBAL) / Instituto Politécnico
13 de Beja (IPBeja), 7801-908 Beja, Portugal

14 ⁴MED – Mediterranean Institute for Agriculture, Environment and Development, CEBAL, 7801-
15 908 Beja, Portugal

16 ⁵Departamento de Biociências, Escola Superior Agrária de Beja, Instituto Politécnico de Beja,
17 Rua Pedro Soares (IPBeja) - Campus do IPBeja, 7800-295 Beja, Portugal

18 ⁶Instituto de Ciências da Terra (ICT), Universidade de Évora, Évora, Portugal

19 ⁷Fiber Materials and Environmental Technologies (FibEnTech), R. Marques de Avila e Bolama,
20 6201-001, Covilhã, Portugal

21
22 Corresponding author e-mail: ppalma@ipbeja.pt

23

1 **ECOTOXICOLOGICAL RISK ASSESSMENT AS A DECISION TOOL FOR**
2 **THE VALORISATION OF WINERY WASTEWATER TREATMENT**
3 **BYPRODUCTS TO AGRICULTURAL PURPOSES**

4
5 Silvana Luz^{1,2,3,4}, Mariana Regato⁵, Alexandra Afonso¹, Adriana Catarino^{1,6}, Teresa Santos¹, Patrícia
6 Palma^{1,6}, Javier Rivas², Fátima Carvalho^{1,7}

7
8 ¹Departamento de Ciências e Tecnologias Aplicadas, Escola Superior Agrária de Beja, Instituto
9 Politécnico de Beja, Rua Pedro Soares (IPBeja) - Campus do IPBeja, 7800-295 Beja, Portugal

10 ²Universidad de Extremadura/Depto Ingeniería Química y química física/University Institute
11 IACYS – Avda. de Elvas, 06006 Badajoz

12 ³Centro de Biotecnologia Agrícola e Agro-Alimentar do Alentejo (CEBAL) / Instituto Politécnico
13 de Beja (IPBeja), 7801-908 Beja, Portugal

14 ⁴MED – Mediterranean Institute for Agriculture, Environment and Development, CEBAL, 7801-
15 908 Beja, Portugal

16 ⁵Departamento de Biociências, Escola Superior Agrária de Beja, Instituto Politécnico de Beja,
17 Rua Pedro Soares (IPBeja) - Campus do IPBeja, 7800-295 Beja, Portugal

18 ⁶Instituto de Ciências da Terra (ICT), Universidade de Évora, Évora, Portugal

19 ⁷Fiber Materials and Environmental Technologies (FibEnTech), R. Marques de Avila e Bolama,
20 6201-001, Covilhã, Portugal

21
22 Corresponding author e-mail: ppalma@ipbeja.pt

23
24
25 **ABSTRACT**

26
27 A simple and economical process consisting in the precipitation with slaked lime (Ca(OH)₂) was
28 applied to winery wastewater. The process not only removed organic matter and other
29 contaminants but also simultaneously captured atmospheric CO₂. In order to help the wine
30 industry to advance to the objective of circular economy, treatment byproducts (supernatant and
31 sludge) have been reused for agricultural purposes due to their physicochemical properties
32 compatible with these applications. In addition to an exhaustive study of the physicochemical
33 characteristics, the ecotoxicological impact of these by-products were also evaluated through the
34 bioindicators: *Thmanocephalus platyurus* (mortality bioassay after 24h), *Vibrio fisheri*

35 (luminescence inhibition after 30 min) and *Pseudokirchneriella subcapitata* (growth inhibition
36 after 72h).

37 Raw winery wastewater showed high toxicity to all bioindicators, being classified as class 4
38 (Toxic Unit (TU); classification proposed by Personne, 1999). According values of TU of 20%
39 for *T. platyurus* (24h), 9.17% for *V. fisheri* (30 min) and 100% for *P. subcapitata* (72h) were
40 found. After treatment (treated WW), the toxicity significantly dropped to class 2, with values of
41 TU of 1.34% for *T. platyurus* (24h), 2.92% for *V. fisheri* (30 min) and 3.97% for *P. subcapitata*
42 (72h).

43 After the immediate one-step lime precipitation process (treated WW), the supernatant was
44 diluted 1:4 (v/v) with groundwater (GW) to generate the so-called effluent nutrient solution
45 (WWNS) which did not present any type of toxicity to the bioindicators used.

46 Accordingly, an assay was carried out to the production of red pak choi, using three different
47 conditions: irrigation with groundwater (control); pH soil correction with organomineral fertilizer
48 and groundwater irrigation; and irrigation with wastewater nutrient solution (WWNS). After the
49 assay none of the soils showed any type of indirect ecotoxicity.

50 The pigment of the red pak choi cultivar produced in this study was evaluated through the analysis
51 of chlorophyll a, chlorophyll b and carotenoids, showing the benefits to the healthy development
52 of the plant.

53

54 **Keywords:** Wine industry; ecotoxicological assays; wastewater industry; slaked lime.

55

56 1. INTRODUCTION

57

58 The Mediterranean area is one of the most important regions in wine production worldwide (Amor
59 et al. 2019; Oliveira et al. 2019), with the European countries representing 61% of total world
60 production (Bolzonella et al. 2019). Portugal is the 11th wine producer in the world, with 7.4
61 million hL in 2021 (IVV IP, 2022), being Alentejo (one of the 14 wine regions, located in the
62 South of Portugal) the 3rd wine producing region of the country, with almost 1.2×10^6 hL of wine
63 produced in 2021/2022 (IVV IP, 2022). In the Alentejo region, the vineyards crops are irrigated,
64 due to the meteorological conditions of the area with low precipitation and high temperatures
65 (Costa et al. 2020).

66 Because of climate change and water scarcity, growth of the wine industry can be questioned
67 (Bolzonella et al. 2019). Reuse of resources, namely water, is of paramount importance to the
68 sustainability of this activity. During the winemaking process (from the reception of the grape to
69 bottling), many resources are used, mainly energy and water (cleaning tanks and barrels, washing

70 equipment and floors) (Amor et al. 2019). In the case of water, consumption can raise up to 450
71 L of water per liter of wine produced (Costa et al. 2020). In addition to the water used in the
72 process, effluents generated are of major concern for the sector, with high variation in
73 composition and flow generated throughout the natural year (Ngwenya et al. 2022). Additionally,
74 the effluent generated contains a wide diversity of substances such as grape films and seeds,
75 yeasts, alcohols (methanol, ethanol and glycerol), soluble acids (acetic, tartaric), tannins, lignin's,
76 polyphenols, residues and cleaning and disinfection products (Ngwenya et al. 2022; Beer et al.
77 2017; Solís et al. 2017). In general, winery effluents present high values of organic matter
78 measured as chemical oxygen demand (COD) 31.37 - 38.39 g L⁻¹, total suspended solids (TSS)
79 34.90 - 76.60 g L⁻¹, as well as high turbidity 319-782 NTU (Ioannou et al. 2015). Due to the
80 seasonal character of this crop, the pH of the effluent varies throughout the year. Additionally,
81 this effluent is characterized by low levels of nutrients such as nitrogen and phosphorus
82 (Bolzonella et al. 2019) and high levels of sodium and potassium (Arienzo et al. 2009). Winery
83 effluents are frequently discharged into the environment or into the sewer system without an
84 adequate pre-treatment stage (Calheiros et al. 2018), presenting a strong environmental impact
85 (Amor et al. 2019). If properly treated, winery wastewater can be reused for agricultural irrigation
86 (Mosse et al. 2012). Winery wastewater treatment studies have already been presented in the
87 specialized literature, such as, coagulation-flocculation (Braz et al. 2010; Rizzo et al. 2010);
88 coagulation-flocculation-decantation and ozonation process (Jorge et al. 2021a) membrane
89 processes (nanofiltration and reverse osmosis) (Ferrarini et al. 2001; Ioannou et al. 2013a);
90 ozonation (Beltrán et al. 1999; Lucas et al. 2009a, 2009b); combination of ozone and radiation
91 and/or hydrogen peroxide (Lucas et al. 2009 a); photo-Fenton (Mosteo et al. 2006; Ormad et al.
92 2006); photo-Fenton with solar radiation (Monteagudo et al. 2012); wet oxidation (Domínguez et
93 al. 2014); anodic oxidation (Lauzurique et al. 2022; Candia-Onfray et al. 2018), anaerobic
94 digestion (Kaira et al. 2022), combination of adsorption and thermocatalytic processes (Jorge et
95 al. 2021b) advanced oxidation processes based on sulfate radicals (Amor et al. 2019),
96 photocatalytic/ photolytic processes (Agustina et al. 2008) and microalgae and immobilized
97 TiO₂/UV-A LEDs (Marchão et al. 2021). However, the application of alkaline precipitants for the
98 treatment of wastewater in wineries still needs to be further studied. Luz et al. 2021, developed a
99 simple and cheap treatment process for winery wastewater which consists of a single precipitation
100 step with slaked lime (Ca(OH)₂) to remove organic matter and other contaminants with
101 simultaneous atmospheric CO₂ capture. That application results in an abundant and insoluble
102 precipitate, capable of sweeping and removing contaminants and the instantaneous combination
103 of calcium and magnesium salts, through the addition of an excess of calcium ions (Luz et al.
104 2021). In addition to the process studied and to the analysis of chemical parameters, an integral
105 evaluation of the treatment efficiency in terms of reuse or disposal is recommended. This process

106 could help the winery industry to achieve circular economy through the reuse of generated
 107 residues (supernatant and sludge).
 108 Application of byproducts for agricultural purposes requires a previous ecotoxicological analysis.
 109 The assessment of ecotoxicological endpoints allow a preliminary analysis of the environmental
 110 impact of these wastewaters (Mendonça et al. 2009). In this line, Sousa et al. (2019) studied the
 111 impact of vinasses wastewater obtained from the distillation of winemaking by-products on
 112 several species, such as *Aliivibrio fisheri*, *Daphnia magna* and *Zebrafish sp.* The acute results
 113 revealed a high toxicity for all species tested, indicating a strong impact in aquatic ecosystem (see
 114 Table 1).
 115 Arienzo et al. (2009) used Garden cress (*Lepidium sativum*) and Onion (*Allium coepa*) in
 116 phytotoxicity bioassays to determine the quality of the winery wastewater (see Table 1). The
 117 authors claimed again a high toxicity, inhibiting the development and growth of the species under
 118 study correlated to high levels of COD, phenols and salinity, and low pH.
 119 Ioannou et al. (2013b) carried out the treatment of winery wastewater by a membrane bioreactor
 120 (MBR) and Fenton's solar oxidation processes. The complete process generated an effluent
 121 without toxicity to the aquatic crustacean *D. magna*, as well as without phytotoxicity (Table 1).
 122 In summary, the main objectives of this study were: (i) to assess the environmental impact of
 123 wastewater from wineries; (ii) analyze the potential reuse of the supernatant and precipitate in
 124 food production.

125
 126 **Table 1.** Ecotoxicological effects of raw and treated winery wastewater

| TYPE OF WASTEWATER | BIOINDICATORS | BIOASSAY | ECOTOXICOLOGICAL ENDPOINTS (EC ₅₀) | STUDY |
|---|--------------------------|-------------------------|--|-------------------------|
| Raw winery | <i>D. magna</i> | Mortality/ immobility | 93.3±2% (24h) | Ioannou et al. 2013b |
| Winery MBR + Fenton solar oxidation process | <i>D. magna</i> | Mortality/ immobility | 26.7±1.5% (24h) | |
| Raw vinasse | <i>D. magna</i> | Mortality/ immobility | 4.8% (48 h) | Sousa et al. 2019 |
| | <i>Zebrafish</i> embryos | Mortality | 0.34% (96 h) | |
| | <i>A. fisheri</i> | Luminescence inhibition | 7.0% (30 min) | |
| Raw winery | <i>Lepidium sativum</i> | Germination | 2.5-15% | Arienzo et al. 2009 |
| | <i>Allium coepa</i> | Germination | 0.25-2% | |

127 EC₅₀ - concentration with effect in 50% of the exposed population.

128

129

130 **2. MATERIALS AND METHODS**

131 This article is a complement of the article of Luz et al. (2021), where a study of the best efficient
132 processes of the immediate lime precipitation and carbonatation with atmospheric CO₂ process
133 was reported. This work adds the description of the preparation of the nutritive solution based on
134 the wastewater from the treated wine cellar and all the ecotoxicological characterization of the
135 different waters, to be applied in agricultural, with correction soil properties and growth of the
136 Red Pak Choi (*Brassica rapa L. var. chinensis (L.) Kitam*), through irrigation with diluted
137 supernatant and incorporation of sludge into the soil.

138

139 **2.1. Winery Wastewater**

140

141 Winery wastewater was collected from a winery located at Vidigueira (Beja, Portugal), that has
142 an integrated production regime, consisting of more than 100 ha of vineyards in sustainable
143 planning, with about 10 grape varieties (Antão Vaz, Síría, Verdelho, Alvarinho, Arinto, Chenin
144 Blanc, Aragonez, Touriga Nacional, Alicante Bouschet and Tinta Miuda).

145 The sampling was carried out in a single day in September 2017, at the winery effluent treatment
146 station in the wastewater (WW) storage tank. Wastewater samples were collected in 5 L
147 polyethylene (PET) packages, subjected to refrigeration at 4°C or freezing at -20°C (the specific
148 procedure was detailed previously in Luz et al. 2021), until analysis.

149

150 **2.2. Soil Characterization**

151

152 Soil samples were collected from a vineyard, whose main characteristic is its acidic nature. After
153 collection, soil samples were air-dried and sieved through a 2 mm sieve. Physicochemical
154 characterization was carried out in triplicate. Extractable P and K were determined using the
155 Egnere-Riehm Method (Riehm, 1958).

156 After air-drying, soil samples were subjected to a water leaching test at room temperature for 24
157 h, under constant stirring, using deionized water to form a solid-liquid solution 1:5 (w/v) ratio.

158 The leachate was separated by centrifugation and filtered through a membrane filter of 0.45 µm
159 pore size. Leachate was analyzed for pH, Electrical Conductivity (EC; dS m⁻¹) and ecotoxicological
160 bioassays.

161

162 **2.3. Metals analysis**

163

164 Metal ions Pb, Cd, Zn, Fe, Ni, Mn, Cu, Ca and Mg were determined in the wastewater samples
165 and leached from the soil, by atomic absorption spectrometry (AAS) after digestion of the samples
166 with aqua regia, according to ISO 11466 (1995). Samples were digested with a mixture of HCl
167 (37%) and HNO₃ (70%), in a ratio 3:1 (v/v), at room temperature for 16 h, and then at 130 ° C for
168 2 h, under reflux conditions.

169 Each suspension was then filtered and diluted to 100 mL with 0.5 mol L⁻¹ HNO₃. Three
170 independent replicates were analyzed for each sample and blanks were measured in parallel.

171

172 **2.4. Ecotoxicological characterization**

173

174 A battery of bioassays was performed in the following samples: Raw Wastewater (WW); Treated
175 WW (TWW); Groundwater (GW); WW nutrient solution (WWNS) (see Table 2) assessing lethal
176 and sub-lethal endpoints, with representative species of key taxonomic and functional groups: i)
177 lethal effects assessed in crustaceans; and ii) two sublethal effects, light and growth inhibition
178 assessed in bacterial and algae.

179 Raw WW is water directly collected from the winery's treatment plant; treated WW is the water
180 treated through the immediate one-step lime precipitation process ((optimal dosage of 5 g L⁻¹ of
181 slaked lime (Luz et al. 2021)), and GW is the water that will be further used to dilute the TWW;
182 WWNS is the combination TWW and GW (dilution 1:4), in order to obtain the optimum pH and
183 EC for plant growth.

184 All bioassays considered have been largely used in ecotoxicological studies and, most of them,
185 have already been standardized. The bioassays were: i) 30 min of luminescence inhibition of *V.*
186 *fischeri* (ISO 11348-2, 1998); ii) 24-h mortality test with *T. platyurus* (Persoone, 1999); iii) 72-h
187 population growth of the green microalgae *P. subcapitata* (OECD 201, 2006). All reported
188 ecotoxicological tests fulfilled the validity requirements established by the respective guidelines.

189

190 **2.4.1. *V. fischeri* bioassay**

191

192 The Luminotox® bioassay was completed by using a marine bacterium *Vibrio fischeri* (NRRL
193 B-11177). Luminescence inhibition was monitored with bacteria supplied as a dry liquid solution
194 (Dr. Lange GmbH & Co. KG, Düsseldorf, Germany, ISO 11348-2 standard (1998) and protocol
195 “DR LANGE luminescent bacteria test”). The solution was stored at -20 °C and rehydrated before
196 tests. These tests were carried out with two replicates of 100% (v/v) of the samples and successive
197 dilutions of 2% NaCl at 50, 25, 12.5, 6.25 and 3.125% (v/v) and non-toxic control of 2% NaCl
198 solution at a fixed temperature of 15±0.5°C. In each sample, bioluminescence was measured

199 before and after the desired incubation period (30 min.) and EC₅₀ (%; the concentration of each
200 sample that reduced 50% of bacterial luminescence) was determined. The decrease in
201 luminescence is due to the inhibition of the bacterial luciferase enzyme by exposure to the
202 samples.

203

204

205 **2.4.2. *T. platyurus* bioassay**

206

207 *The Thamnocephalus platyurus* bioassay was carried out following the protocol used in the
208 THAMNOTOXKIT F™ kit (Persoone, 1999). This test evaluates the mortality of shrimp larvae
209 (<24h) incubated in 24-well plates, with 1.0 mL of the test solution and ten crustaceans per well.
210 Four repetitions per treatment were completed at 25 °C for 24 hours in the absence of light, without
211 feeding the animals during the time of the test. Concentrations of 12.5, 25, 50, 100% (v/v) were
212 tested using the samples of supernatant. And concentrations of 25, 50, 100% (v/v) were tested
213 using the samples (included in the test kit, also used as a non-toxic control) for the soils. The 24
214 hour contact with each test solution was used as the selected end point, and the EC₅₀ (%) was
215 calculated.

216

217 **2.4.3. Growth inhibition of green microalgae *P. subcapitata***

218

219 The determination of the fluorospectrometric growth rate of the unicellular microalgae *P.*
220 *subcapitata* was carried out for 72 hours, (protocol of OECD 201 (2006)), with the MBL Medium
221 Woods-Holecontrol medium. At the beginning of the test, *P. subcapitata* (100 µL of an inoculum
222 with 3-5×10⁴ cells mL⁻¹) was exposed to water samples (900 µL) in a 24-well microplate. The
223 test flasks were randomly incubated in an orbital shaker for 72 hours at a constant temperature of
224 21±2°C and with a constant light intensity (60–120 µE m⁻² s⁻¹, equivalent to 6.000–10.000 lx).
225 Concentrations were calculated using the microscope and Neubauer chamber.

226 The average rate of specific growth for a specific period was calculated as the logarithmic increase
227 in biomass (after 72 hours) using the equation:

228

$$229 \mu_{i-j} = (\ln B_j - \ln B_i) / (t_j - t_i) \quad (1)$$

230

231 Where: μ_{i-j} is the average specific growth rate from time i to j; t_i is the time for the start of the
232 exposure period; t_j is the time for the end of the exposure period, B_i is the biomass concentration
233 at time i, and B_j is the biomass concentration at time j.

234 The inhibition of algae growth was estimated as the percentage of reduction in the growth rate
235 compared to the control:

236

$$237 \quad \% I = [(\mu_c - \mu_t) / \mu_c] \times 100 \quad (2)$$

238

239 Where: % I is the mean percentage of inhibition for specific growth rate; μ_c is the mean value for
240 the growth rate in the control, and μ_t is the mean value for the growth rate in water samples.

241

242 **2.5. Red Pak Choi cultivation. Experimental design**

243

244 Red Pak Choi was grown in pots with 1.5 kg of acidic soil under three different conditions: i)
245 irrigated with groundwater (soil control); ii) correction of soil pH with organomineral fertilizer +
246 underground irrigation; iii) irrigation with wastewater nutrient solution (WWNS). The
247 experimental design was randomized with nine replications, and the experiment was conducted
248 outdoors under uncontrolled temperature conditions, from May to July. The plants were placed
249 in the tests when they had only cotyledonary leaves.

250 The studied acid soil had a pH of 5.3 ± 0.1 , K $152.7 \pm 4.2 \text{ mg L}^{-1}$ K₂O fine soil, P $32.9 \pm 3.6 \text{ mg L}^{-1}$
251 P₂O₅ fine soil classified as sandy loam, $0.8 \pm 0.1\%$ organic matter, with 50.6% coarse sand, 30.1%
252 fine sand, 8.8% silt and 10.2% clay (Table 4).

253 Monitoring of plant growth was through marking and measurements. Leaf biomass was sampled
254 for determination of moisture, chlorophylls (a, b) and carotenoids.

255

256 **2.6. Preparation of WW nutrient solution**

257

258 WWNS was prepared from the TWW, with the addition of GW, until the ideal pH and
259 conductivity characteristics for plants growth (*Brassica rapa L. var. chinensis (L.) Kitam*) were
260 obtained. These vegetables are considered moderately salinity sensitive with a threshold EC of
261 1.3 dS m^{-1} , and the salinity of the irrigation water can compromise the vegetative growth and root
262 development of the plant. (Ünlükara et al. 2008).

263 Dilution with GW was optimized to achieve the reduction of water salinity (see Table 2) as well
264 as its toxic effect. According to Figure 1, toxicity is eliminated after dilution above 25%. Thus,
265 the ideal condition for WWNS is the addition of 1:4 groundwater, obtaining a pH of 7.2 ± 0.13 and
266 EC of $1.1 \pm 0.63 \text{ dS m}^{-1}$ (Table 2).

267

268

269

270

271

272 **Table 2.** Characterization of WW Samples

| | Units | Raw WW* | TWW* | GW | WWNS |
|--------------------------------------|--|----------------------|----------------|-------------|--------------|
| Parameter | | | | | |
| pH | Sorensen scale | 4.26±0.03 | 12.41±0.24 | 7.79±0.05 | 7.17±0.13 |
| Electric | dS m ⁻¹ | 1.243±0.034 | 6.545±0.62 | 0.525±0.16 | 1.140±0.63 |
| Conductivity | | | | | |
| Total hardness | mg CaCO ₃ L ⁻¹ | 424.78±53.10 | 1804.97±350.75 | 319.17±5.52 | 345.69±23.77 |
| Magnesium | mg Mg L ⁻¹ | 98.71±13.33 | 179.89±58.63 | 32.01±2.07 | 76.75±1.86 |
| Calcium | Mg Ca L ⁻¹ | 7.09±6.14 | 426.76±84.46 | 75.12±5.35 | 134.65±8.59 |
| COD | mg O ₂ L ⁻¹ | 3225±25 | 2509±166 | 0 | 448±48 |
| BOD₅ | mg O ₂ L ⁻¹ | 1266 ±208 | 280±0 | 11±1 | 17 ±2 |
| BOD₅/COD | - | 0.39 | 0.11 | --- | 0.04 |
| Turbidity | NTU | 159.77±7.68 | 2.00±0.84 | --- | --- |
| N- NH₄⁺ | mg NH ₄ L ⁻¹ | 0.72±0.00 | 5.38±0.00 | 0 | 0 |
| N Kjeldhal | mg N L ⁻¹ | 22.3±1.6 | 7.58±0.00 | 0.19±0.32 | 1.86±0.32 |
| Ptotal | mg P L ⁻¹ | 155.69±6.53 | 20.3±8.65 | 4.41±1.78 | 9.54±0 |
| Phosphates | mg PO ₄ L ⁻¹ | --- | --- | --- | --- |
| Chlorides | mg L ⁻¹ | 124.7±1.5 | 121.9±23.8 | 23.3±0.15 | 78.9±5.8 |
| Nitrates | mg NO ₃ ⁻ L ⁻¹ | 5.9±0.4 | 12.6±2.7 | 86.7±0.3 | 21.2±0 |
| Fluorides | mg L ⁻¹ | 166.7±2.3 | 176.0±11.1 | 0.03±0.0 | 36.3±0.1 |
| Sulphates | mg SO ₄ ²⁻ L ⁻¹ | 95.4±0.5 | 56.5±14.3 | 34.4±0.2 | 52.5±0.3 |
| Iron | mg Fe L ⁻¹ | 5.725±1.591 | 0.030±0.009 | --- | 0.030±0.001 |
| Manganese | mg Mn L ⁻¹ | 0.173±0.018 | 0.005±0.007 | --- | 0.005±0.001 |
| <i>E. Coli</i> | ufc/100mL | 0 | 0 | 0 | --- |
| Fecal coliforms | ufc/100mL | 45±29.8 | 0 | 0 | --- |
| Enterococcus | NMP/mL | 3.7×10 ⁻³ | 0 | 0 | --- |

273 * Luz et al. 2021 (mean±SD; n = 3; p<0.05)

274 **2.7. Sludge**

275

276 After applying the processes of immediate precipitation of lime, used in the study, the
 277 sedimentation and decantation process is carried out, where the TWW is separated from the
 278 precipitate (sludge), which, due to its characteristics, pH 12.40±0.02, EC 7.25±0.15 dS m⁻¹,
 279 Magnesium 0.0381±0.0067 mg L⁻¹ (Luz et al. 2021), was used as an agricultural corrective
 280 in the studied acid soil.

281

282

283

284

285 2.8. Statistics

286

287 In the *V. fischeri* bioluminescence inhibition test, the EC₅₀ (%) values were
288 determined using LUMISsoft 4 Software™. The EC₅₀ (%) for the *T. platyurus* and *D.*
289 *magna* mortality were determined using the probit analysis (Finney 1971). Data for the
290 inhibition growth bioassay were checked for homogeneity of variance by the
291 Kolmogorov–Smirnov test and, when possible, subjected to one-way analysis of variance
292 (ANOVA). Data which did not satisfy the assumption for ANOVA, were analyzed non-
293 parametrically using Kruskal–Wallis ANOVA by ranks test. Whenever significant
294 differences were found ($p < 0.05$), a post hoc Dunnett's test was used to compare treatments
295 with the control, for an alpha value of 0.05 (Zar 1996).

296 For a better interpretation of toxicity data, all measurements (EC₅₀) were converted into
297 Toxicity Units (TU), i.e. the inverse of EC₅₀ values, expressed in percent:

$$298 \text{ TU (\%)} = 1/\text{EC}_{50} \times 100 \quad \text{(Equation 5)}$$

299 With the toxicity classification system (TCS) reported by Personne et al. (2003), it is
300 possible to classify the samples as: class 1 (TU < 1) exhibiting no significant toxicity; class 2 (1
301 < TU < 10) exhibiting significant toxicity; class 3 (10 < TU < 100) exhibiting high toxicity; and
302 class 4 (TU > 100) exhibiting very high toxicity.

303 The results of the parameters analyzed in the remaining tests were submitted to analysis of
304 variance (ANOVA), and their means were compared by Tukey's test, at the 0.05% level of
305 significance, using the GraphPad.

306 In the ANOVA analysis of variance, it was determined that whenever the F value was greater
307 than the critical F, there was a significant effect between treatments. Whenever F was less than
308 the critical F, there was no statistically significant difference.

309

310 3. RESULTS

311

312 3.1. Ecotoxicological Assessment

313 This study presents results of the ecotoxicological analysis of several water matrices obtained
314 from the chemical treatment applied to winery wastewaters, and the assessment of their potential
315 use (diluted with groundwater) as irrigation water. The ecotoxicological assessment allowed to

316 analyze the risk to the aquatic ecosystem by using the irrigation solution prepared (produced from
 317 treated winery wastewaters). The ecotoxicological results using the bioindicators *T. platyurus*, *V.*
 318 *fischeri* and *P. subcapitata* are summarized in Table 3. Further, the results for growth inhibition
 319 of *P. subcapitata* (a sub-lethal parameter) are shown in Figure 2.

320 The lethal ecotoxicological results (Toxic Unit (TU); Table 3) indicated that the winery effluent
 321 was very toxic to the three bioindicators, being classified as class 4 (TU > 100; with very high
 322 toxicity), constituting a serious danger to the receiving environment. The green microalgae was
 323 the organism most sensitive to this effluent with a 72h-EC₅₀ = 0.02% for growth inhibition. The
 324 toxicity detected may be correlated to: (i) the concentrations of chlorides (125 mg L⁻¹), sulphates
 325 (195 mg L⁻¹), ammonium (0.72 mg L⁻¹), and fluorides (7 mg L⁻¹) (see Table 2); and (ii) unknown
 326 organic contaminants. Authors such as, Alzahrani et al. (2013), Palma et al. (2009) and Alvarenga
 327 et al. (2009), verified the influence of concentrations of ammonia, chlorides and sulfates in the
 328 growth inhibition or mortality of aquatic organisms. The ecotoxicological effect of the one-step
 329 lime precipitation process, i.e., the capacity to reduce the wastewater ecotoxicity, was evident,
 330 resulting in a treated wastewater classified as class 2 (using the TU classification) with a lower
 331 acute toxicity for all the bioindicators used. The lime precipitation treatment led to the decrease
 332 of the concentrations of chlorides, sulphates, iron and manganese (see table 3). Despite that,
 333 ammonia concentrations are still high 5.4 mg L⁻¹, as well as sulfates concentrations (52.5 mg SO₄²⁻
 334 L⁻¹) and fluorides (36.3 mg L⁻¹), contributing, therefore, to the remaining observed toxicity.
 335 Dilution of the treated wastewater with groundwater (1:4) resulted in a water matrix classified as
 336 class 1 (TU < 1) in terms of toxicity. Although most studies conducted with wastewaters reported
 337 *V. fischeri* as the most sensitive bioindicator to ecotoxicological assays (Alzahrani et al., 2013;
 338 Latorre et al. 2007; Zgórska 2011; Palma et al. 2016), these results showed that *P. subcapitata*
 339 was in this case the most sensitive organism. This fact highlighted the need to include these two
 340 bioindicators in a battery of ecotoxicity tests to assess the environmental risk of industrial
 341 effluents, as well as in the discussion of the effectiveness of different effluent treatments.

342

343 **Table 3.** Values of TU (%) obtained with *V. fischeri*, *P. subcapitata* and the crustacean *T.*
 344 *platyurus* when exposed to the winery wastewater (WW), treated wastewater (TWW);
 345 groundwater (GW) and wastewater nutritive solution (WWNS). The matrices were classified into
 346 toxicity classes according to their value of toxicity units (TU).

| TU (%) | | | | |
|--------|------------------------------|--------------------------------|--------------------------------|---------|
| | <i>T. platyurus</i> (24h) | <i>V. fischeri</i> (30 min) | <i>P. subcapitata</i> (72h) | |
| WW | 20 | 9.17 | 100 | Class 4 |

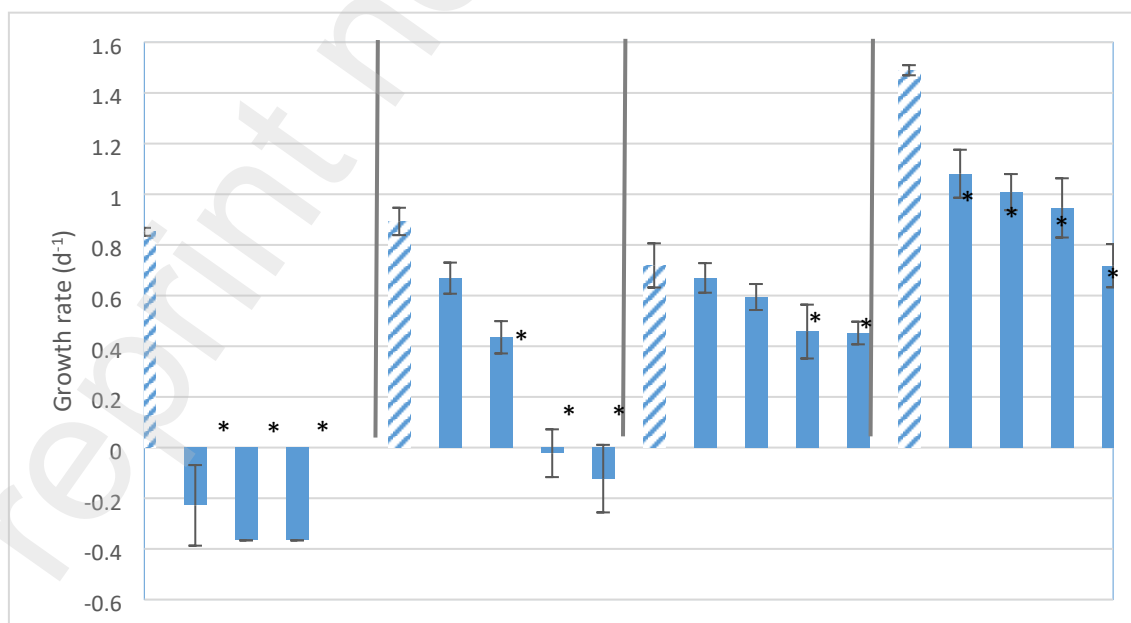
| | | | | |
|-------------------|-----------|-----------|-----------|---------|
| Treated WW | 1.34 | 2.92 | 3.97 | Class 2 |
| GW | non toxic | non toxic | non toxic | Class 1 |
| WWNS | non toxic | non toxic | non toxic | Class 1 |

347

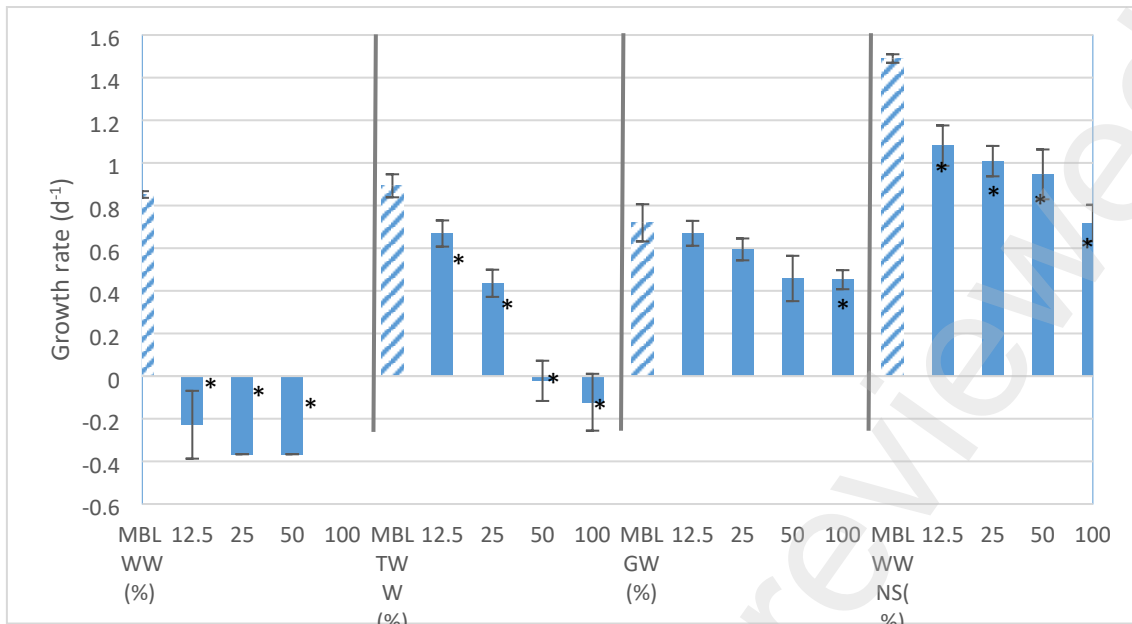
348 The ecotoxicological assessment through sublethal parameters showed that the winery wastewater
 349 significantly induced growth inhibition to the microalgae *P. subcapitata* (one-way ANOVA;
 350 $p < 0.05$), with 100% of inhibition for all the concentrations exposed (Figure 1). The results
 351 indicated that all matrices induced some growth inhibition of microalgae, less accentuated in
 352 groundwater, with growth inhibition rates always below 30% (significantly different to the
 353 control; one-way ANOVA; $F_{4,18} = 10.353$, $p = 0.00016$).

354 WWNS (obtained from treated wastewater diluted with groundwater) induced a growth inhibition
 355 rate of 45%, indicating toxicity for the evaluated sublethal endpoints (significantly different to
 356 the control; one-way ANOVA; $F_{4,20} = 48.423$, $p = 0.00000$). Therefore, despite the decrease in
 357 toxicity, the irrigation solution still exhibited some toxicity to microalgae. The results can be
 358 explained considering the sensitivity of the bioindicator and the toxicological endpoints used
 359 (sublethal bioassays assess different stages of the bioindicator's life cycle, being more sensitive
 360 to lower concentrations of contaminants). In fact, several authors reported toxic effects in the
 361 microalgae *P. subcapitata*, when exposed to low concentrations of organic compounds such as
 362 polycyclic aromatic hydrocarbon (PAHs; Eom et al. 2007), pharmaceuticals (Yang et al., 2008)
 363 and herbicides (Pérez et al. 2011).

364



365



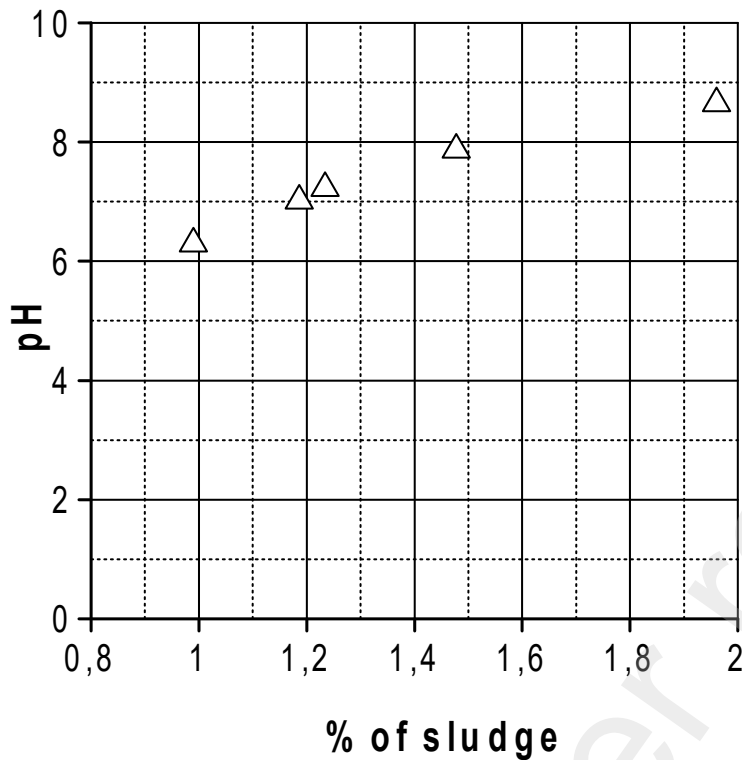
366

367 **Figure 1.** Growth rate (d⁻¹) of the microalgae *P. subcapitata* after 3 days exposed to the winery
 368 wastewater (WW), treated wastewater (TWW); ground water (GW) and Wastewater nutrient
 369 solution (WWNS) (mean±SD; n = 6; * p<0.05, Dunnett's test with a control (MBL)).

370

371 3.2. Use of sludge amended soils in cultivars

372 Soil pH correction was carried out by addition of different amounts of sludge to raw soil. Figure
 373 2 depicts the final pH achieved as a function of the percentage of added sludge. As observed with
 374 just an approximated 1.2% (w/w) of sludge the pH of the soil can be corrected to circumneutral
 375 conditions, more adequate to red pak choi growth.



376

377 **Figure 2.** pH correction of acidic soil as a function of sludge percentage addition.

378

379 Table 4 compares the characterization of raw soil and corrected soil. Apart of pH, main
 380 differences are found in P content increasing from roughly 33 mg g⁻¹ to 64 mg g⁻¹ as P₂O₅ and K
 381 raising from 153 to 168 mg g⁻¹ as K₂O. These results are sustained by the content in phosphorus
 382 and potassium of the obtained sludge after treatment, specifically 895 and 990 mg g⁻¹ in P₂O₅ and
 383 K₂O, respectively (Luz et al. 2021). The critical level of nutrients differs depending on
 384 environmental conditions, soil type, plant type, method of extraction, etc. In any case, broadly
 385 speaking, P in corrected soil (14 mg g⁻¹ as P) seems to be in the range of medium level content
 386 assuming that all P was available for plants while K content (139 mg g⁻¹ as K) is considered
 387 slightly low if all K were extractable. The other remarkable change in concentration corresponds
 388 to Ca, which was expected given the sludge origin. Calcium deficiencies usually are found only
 389 on very acidic soils.

390

391

392

Table 4. Characterization of raw (acidic) and pH corrected (1.2 %) soils

| | Acidic Soil | | Corrected soil | |
|---|-------------|--------|----------------|--------|
| | Mean | SD | Mean | SD |
| pH | 5.31 | 0.12 | 7.23 | 0.08 |
| EC, mS cm ⁻¹ | 0.27 | - | 0.40 | - |
| % TOC | 0.45 | 0.03 | 0.49 | 0.02 |
| % Organic matter | 0.8 | 0.1 | 0.8 | - |
| K, mg g ⁻¹ K ₂ O | 152.7 | 4.2 | 168.1 | 10.5 |
| P, mg g ⁻¹ P ₂ O ₅ | 32.9 | 3.6 | 63.7 | 1.7 |
| Pb, µg g ⁻¹ DW | 32.40 | 1.15 | 34.10 | 1.10 |
| Cd, µg g ⁻¹ DW | 2.20 | 0.20 | 2.20 | 0.50 |
| Zn, mg g ⁻¹ DW | 0.2366 | 0.1744 | 1.0197 | 0.5317 |
| Fe, mg g ⁻¹ DW | 96.647 | 9.1393 | 100.368 | 4.374 |
| Ni, µg g ⁻¹ DW | 77.30 | 5.00 | 73.90 | 2.00 |
| Mn, mg g ⁻¹ DW | 1.5116 | 0.0329 | 1.4345 | 0.0659 |
| Cu, mg g ⁻¹ DW | 0.2283 | 0.0118 | 0.2207 | 0.0088 |
| Ca, mg g ⁻¹ DW | 10.013 | 0.6908 | 102.00 | 20.274 |
| Mg, mg g ⁻¹ DW | 13.677 | 0.3470 | 15.011 | 0.6881 |

394 DW = Dry weight

395

396 The ecotoxicological evaluation of the different soil leachates (acid soil, soil irrigated with
 397 WWNS and corrected soil), allowed the analysis of the risk to the ecosystem after the application
 398 of the prepared irrigation solution (WWNS) and the incorporation of the precipitate as a way of
 399 correcting the pH of the soil. ground. Ecotoxicological results using the bioindicators *T. platyurus*,
 400 *V. fisheri* are summarized in Table 5 and the bioindicator *P. subcapitata* in Figure 3. Lethal
 401 ecotoxicological results (Toxic Unit (TU); Table 5) indicated that both irrigation and correction
 402 did not induce the increment of soil toxicity.

403

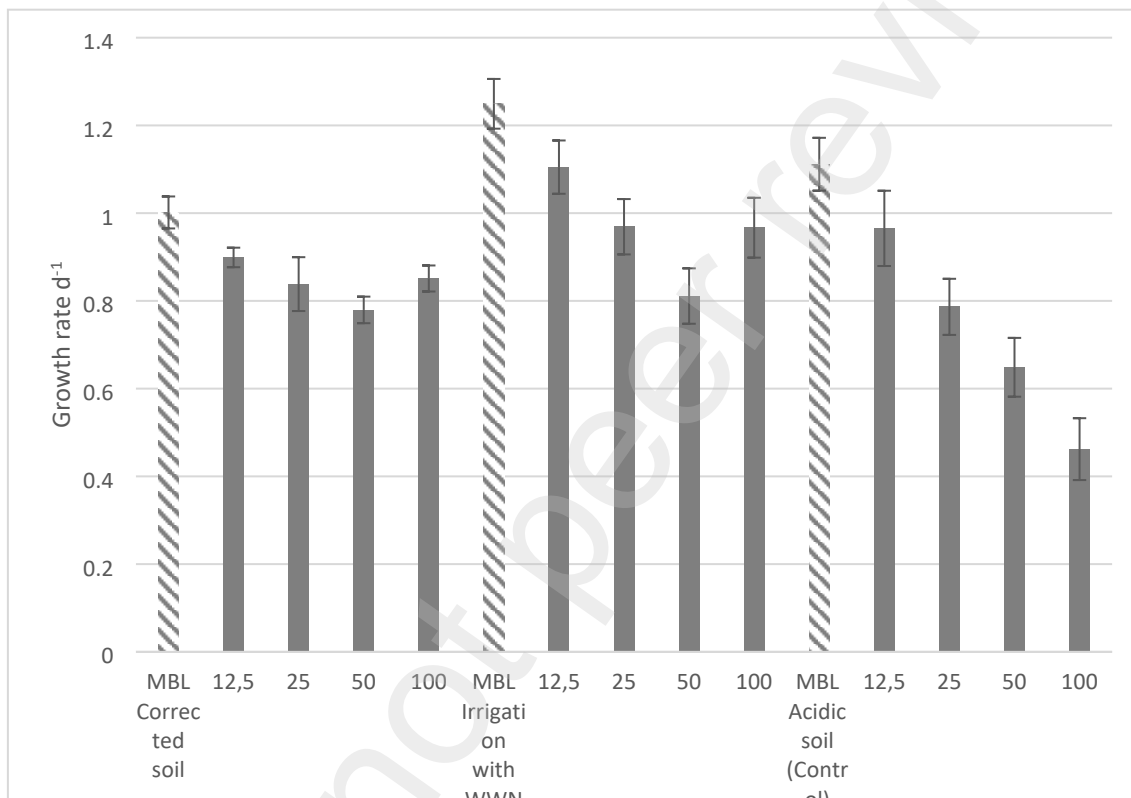
404 **Table 5.** Values of TU (%) obtained with *V. fisheri*, and the crustacean *T. platyurus* when
 405 exposed to the soils, acidic soil (control); irrigation with WWNS and corrected soil. The matrices
 406 were classified into toxicity classes according to their value of toxicity units (TU).

| TU (%) | |
|---------------------|-------------------|
| <i>T. Platyurus</i> | <i>V. fisheri</i> |
| | |

| | (24h) | (30 min) | |
|------------------------------|-----------|-----------|---------|
| Acidic soil (Control) | non toxic | non toxic | Class 1 |
| Irrigation with WWNS | non toxic | non toxic | Class 1 |
| Corrected soil | non toxic | non toxic | Class 1 |

407

408



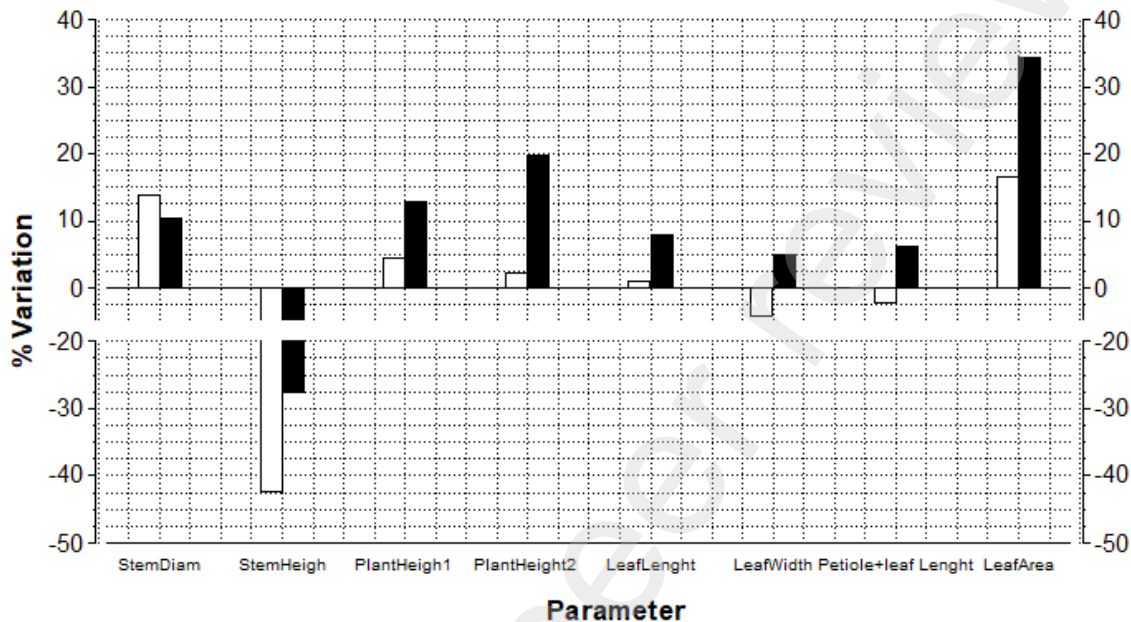
409

410 **Figure 3.** Growth rate (d⁻¹) of the microalgae *P. subcapitata* after 3 days exposed to the corrected
 411 soil, irrigation with WWNS and acidic soil (mean±SD; n = 6; * *p*<0.05, Dunnett's test with a
 412 control (MBL).

413

414 Once soils were characterized, the potential benefits of soil sludge amendment were assessed by
 415 analyzing the evolution of a control cultivar (red pak choi) under three different scenarios. Hence,
 416 red pak choi was grown in the raw acidic soil by using either groundwater or treated wastewater
 417 in the irrigation. The wastewater used was obtained after dilution (1:4) and neutralization of
 418 treated wastewater with groundwater (Luz, 2021).

419 Several parameters were measured to compare the effect of soil amendment and water irrigation
 420 nature on red pak choi (see Figure 4). Hence, regarding plant size a positive effect of soil
 421 correction and use of wastewater was experienced in stem diameter, plant height and leaf
 422 length/area while a decrease in stem height was obtained when the plants were cultivated under
 423 the studied systems if compared to values in acidic soil.



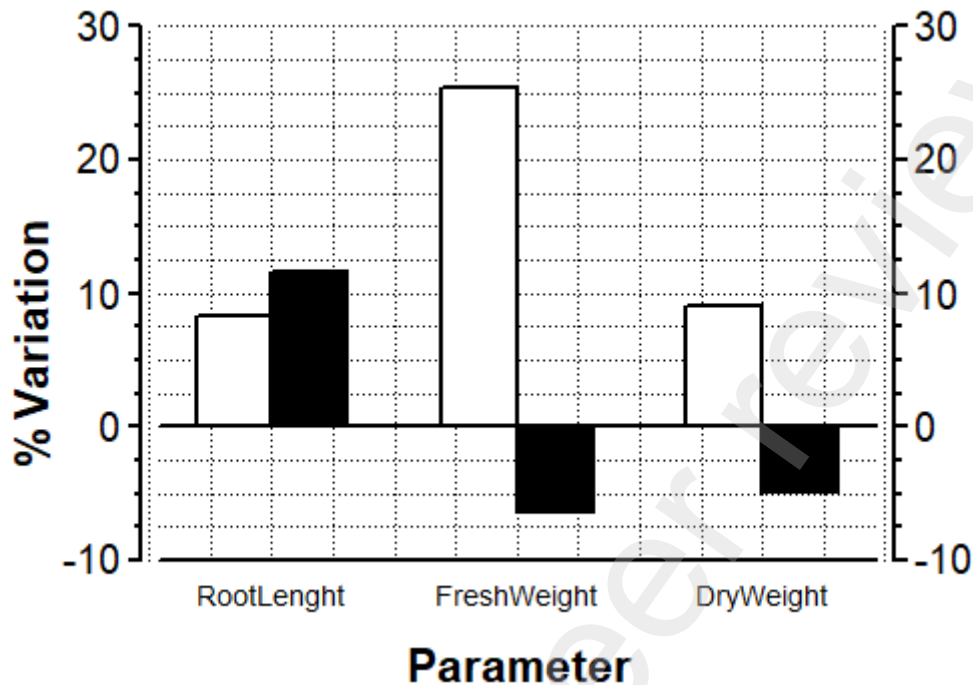
424

425 **Figure 4.** Use of irrigation with WWNS on acid soil (white bars) and irrigation with GW on
 426 amended soil (black bar) on red pak choi growth. Parameters compared to results obtained in
 427 uncorrected acidic soil irrigated with GW (PlantHeigh1 measured from ground to the longest leaf
 428 stretched, PlantHeigh2 measured from ground to the largest leaf without stretching).

429 Broadly speaking, cultivars grown in amended soil (irrigated with GW) presented better
 430 characteristics than those produced by irrigating with WWNS the uncorrected soil. Moreover,
 431 wastewater watering led to a reduction in leaf width and petiole+leaf length if compared to control
 432 experiments. The most remarkable differences in amended soil were found in plant height and
 433 leaf area with an increase of roughly 20% (PlantHeigh2) and 34%, respectively, contrasted to
 434 results obtained in the acidic soil. The only parameter negatively affected by soil amendment was
 435 stem height with a reduction of 28% (see Figure 5).

436 The higher size of plants was translated to a notable increase in the dry weight of the aerial part
 437 of the vegetables, experiencing a 23% and 42 % raise in plants grown with WWNS and corrected
 438 soil, respectively.

439 Contrarily to aerial data, use of corrected soil led to a decrease of root weight although an increase
440 in length was experienced. Watering with WWNS increased the size and weight of roots,
441 especially fresh weight with a raise of 25%.



442

443 **Figure 5.** Use of irrigation with WWNS (white bars) or amended soil (black bar) on red pak
444 choi growth. Root parameters compared to results obtained in uncorrected acidic soil.

445

446 3.3. Pigment content in red pak choi grown

447 Chlorophyll is the designation of a group of photosynthetic pigments found in the chloroplasts of
448 plants. In green plants, chlorophyll appears in two forms: chlorophyll a and b.

449 This pigment, together with carotenoids, has the ability to capture solar radiation and allow plants
450 to carry out photosynthesis, a vital function of plants, which allows the production of
451 carbohydrates, which they use to produce essential energy for their development. and
452 productivity.

453 During photosynthesis, CO₂ is consumed, and oxygen is released, hence the importance of plants
454 for maintaining oxygen and sequestering carbon on the planet (Moreira, 2013)

455 As can be seen in table 6, the plants watered with WWNS, as well as those grown in the soil,
456 where the pH correction was carried out, had higher levels of chlorophyll, which is a positive
457 aspect, since they are plants with higher development capacity and greater productive capacity.

458 Nitrogen is part of chlorophyll and is directly involved in photosynthesis (Rodrigues and
459 Coutinho, 2000).

460 The content in pigments (chlorophyll and carotenoids) can be an indirect tool to assess nitrogen
461 availability in soils. Photosynthetic pigments can, therefore, detect deficiencies in nitrogen
462 (Balasu-Bramanian et al., 2000; Torres Netto et al., 2005). As a rule of thumb, deficiencies in N,
463 P and K affect photosynthesis and growth (Muñoz Huerta et al., 2013), and, chlorophyll and
464 carotenoids content is a valuable tool to reveal the shortage of these essential elements (Sanchez
465 et al., 2018). As inferred from table xx, cultivars grown with irrigation with WWNS or in amended
466 soils present higher levels of chlorophyll than those obtained in acidic soil. Carotenoids content
467 does not significantly vary with the procedure of cultivation. These results suggest a positive
468 influence of TWW and sludge amendment on NPK availability to plants.

469

470 **Table 6.** Pigment content in red pak choi grown under three scenarios

| | Chlorophyll a mg g ⁻¹ | | Chlorophyll b mg g ⁻¹ | | Carotenoids mg g ⁻¹ | |
|--------------------------|-------------------------------------|-------|-------------------------------------|-------|-----------------------------------|-------|
| | Mean | SD | Mean | SD | Mean | SD |
| Acidic soil (control) | 1.55 | 0.265 | 1.06 | 0.262 | 0.24 | 0.040 |
| Irrigation with WWNS | 1.925 | 0.095 | 1.53 | 0.235 | 0.264 | 0.081 |
| Corrected soil | 1.845 | 0.178 | 1.38 | 0.123 | 0.25 | 0.106 |

471

472

473 **4. CONCLUSION**

474

475 Winery wastewater represents a serious environmental problem when not treated. Since the
476 annual production of these tributaries is very significant, their use can be a solution for the future,
477 offering profitability and promoting a circular economy for this industry. The raw effluent (WW)
478 showed high toxicity, which decreased significantly after the application of the slaked lime
479 treatment. To use it for irrigation, it was essential to meet the pH and conductivity characteristics
480 that the plant needs, being necessary to dilute the TWW with GW, thus making it an ideal nutrient

481 solution for the cultivar. Ecotoxicologically, WWNS did not show any toxicity. An agronomic
482 study was then carried out using three different conditions: irrigation with groundwater (soil
483 control); correction of soil pH with organomineral fertilizer and underground water irrigation;
484 irrigation with WWNS, for the production of red pak choi.

485 All soils were analysed ecotoxicologically using indirect bioassays, and none showed toxicity.
486 After the cultivar reached the optimal harvesting conditions, it was analysed in terms of
487 pigmentation, through the study of chlorophylls a, b and carotenoids. All parameters presented
488 favourable conditions for healthy plant growth.

489

490 **DECLARATIONS**

491

492 **Consent for publication**

493

494 All authors agreed with the content and that all gave explicit consent to submit.

495

496 **Acknowledgement**

497

498 The authors want to thank the FCT - Foundation for Science and Technology for the PhD
499 scholarship awarded to Silvana Luz (SFRH/BD/129849/2017).

500

501 **2. References**

502

503 Agustina, T.E., Ang, H.M., Pareek, V.K., 2008. Treatment of winery wastewater using a
504 photocatalytic/photolytic reactor. *Chemical Engineering Journal* 135, 151-156.
505 <https://doi.org/10.1016/j.cej.2007.07.063>

506 Alvarenga, P., Palma, P., Gonçalves, A.P., Fernandes, R.M., Varennes, A., Vallini, G., Duarte,
507 E., Cunha-Queda, A.C., 2009. Organic residues as immobilizing agentes in aided
508 phytostabilization: (II) Effects on soil biochemical and ecotoxicological characteristics.
509 *Chemosphere* 74, 1301-1308. <https://doi.org/10.1016/j.chemosphere.2008.11.006>

510 Alzahrani, S., Mohammad, A.W., Hilal, N., Abdullah, P., Jaafar, O., 2013. Comparative study of
511 NF and RO membranes in the treatment of produced water II: Toxicity removal efficiency.
512 *Desalination* 315, 18-26. <https://doi.org/10.1016/j.desal.2012.12.004>

513 Amor, C., Rodríguez-Chueca, J., Fernandes, J.L., Domínguez, J.R., Lucas, M.S., Peres, J.A.,
514 2019. Winery wastewater treatment by sulphate radical based-advanced oxidation processes (SR-
515 AOP): Thermally vs UV-assisted persulphate activation. *Process Safety and Environmental*
516 *Protection* 122, 94–101. <https://doi.org/10.1016/j.psep.2018.11.016>

517 Arienzo, M., Christen, E., Quayle, W., 2009. Phytotoxicity testing of winery wastewater for
518 constructed wetland treatment. *Journal of Hazardous Materials* 169, 94–99.
519 <https://doi.org/10.1016/j.jhazmat.2009.03.069>

520 Balasubramanian, V., Morales, A.C., Cruz, R.T., Thiyagarajan, T.M., Nagarajan, R., Babu, M.,
521 Abdulrachman, S., Hai, L.H., 2000. Adaptation of the chlorophyll meter (SPAD) technology for
522 real-time N management in rice: a review. *International Rice Research Notes*, 25, 4-8.
523 <https://doi.org/10.5281/zenodo.7001703>

524 Beer, D.M., Botes, M., Cloete, T.E., 2017. The microbial community of a biofilm contact reactor
525 for the treatment of winery wastewater. *Journal of Applied Microbiology* 124, 598-609.
526 <https://doi.org/10.1111/jam.13654>

527 Beltrán, F.J., García-Araya, J.F., Álvarez, P.M., 1999. Wine Distillery Wastewater Degradation.
528 1. Oxidative Treatment Using Ozone and Its Effect on the Wastewater Biodegradability. *J. Agric.*
529 *Food Chem*, 47, 9, 3911–3918. <https://doi.org/10.1021/jf981262b>

530 Bolzonella, D., Papa, M., Da Ros, C., Muthukumar, L.A., Rosso, D., 2019. Winery wastewater
531 treatment: a critical overview of advanced biological processes. *Critical Reviews in*
532 *Biotechnology*. <https://doi.org/10.1080/07388551.2019.1573799>

533 Braz, R., Pirra, A., Lucas, M.S., Peres, J.A., 2010. Combination of long term aerated storage and
534 chemical coagulation/flocculation to winery wastewater treatment. *Desalination* 263, 226- 232.
535 <https://doi.org/10.1016/j.desal.2010.06.063>

536 Calheiros, C.S.C., Pereira, S.I.A., Castro, P.M.L., 2018. Culturable bacteria associated to the
537 rhizosphere and tissues of *Iris pseudacorus* plants growing in a treatment wetland for winery
538 wastewater discharge. *Ecological Engineering*, 115, 67-74.
539 <https://doi.org/10.1016/j.ecoleng.2018.02.011>.

540 Candia-Onfray, C., Espinoza, N., Silva, E.B.S., Toledo-Neira, C., Espinoza, L.C., Santander, R.,
541 García, V., Salazar, R., 2018. Treatment of winery wastewater by anodic oxidation using BDD
542 electrode. *Chemosphere* 206, 709-717. <https://doi.org/10.1016/j.chemosphere.2018.04.175>

543 Costa, J.M., Oliveira, M., Egipto, R.J., Cid, J.F., Fragoso, R.A., Lopes, C.M., Duarte, E.N., 2020.
544 Water and wastewater management for sustainable viticulture and oenology in South Portugal –
545 a review. *Ciência Téc. Vitiv.* 35(1), 1-15. <https://doi.org/10.1051/ctv/20203501001>

546 Domínguez, C.M., Quintanilla, A., Casas, J.A., Rodriguez, J.J., 2014. Treatment of real winery
547 wastewater by wet oxidation at mild temperature. *Sep. Purif. Technol.*, 129, 121-128.
548 <http://dx.doi.org/10.1016/j.seppur.2014.04.003>

549 Eom, I.C., Rast, C., Veber, A.M., 2007. Ecotoxicology of a polycyclic aromatic hydrocarbon
550 (PAH) – Contaminated soil. *Ecotoxicology and Environmental Safety* 67, 113-116.
551 <https://doi.org/10.1016/j.ecoenv.2006.12.020>

552 Ferrarini, R., Versari, A., Galassi, S., 2001. A preliminar comparison between nanofiltration and
553 reverse osmosis membranes for grape juice treatment. *Journal of food engineering* 50:2, 113-116.
554 [https://doi.org/10.1016/S0260-8774\(00\)00199-0](https://doi.org/10.1016/S0260-8774(00)00199-0)

555 Finney, D.J., 1971. *Probit Analysis* Cambridge, UK Cambridge University Press.

556 Instituto da Vinha e do Vinho – IVV, 2022. Evolução da Produção Nacional de Vinho por Região
557 Vitivinícola Série 2009/2010 a 2021/2022. <https://www.ivv.gov.pt/np4/163.html>. Accessed 02
558 December 2022.

559 Ioannou, L.A., Michael, C., Kyriakou, S., Fatta-Kassinos, D., 2013b. Solar Fenton: from pilot to
560 industrial scale application for polishing winery wastewater pretreated by MBR. *J Chem Technol*
561 *Biotechnol* 89, 1067-1076. <https://doi.org/10.1002/jctb.4203>

562 Ioannou, L.A., Michael, C., Vakondios, N., Drosou, K., Xekoukoulotakis, N.P., Diamadopoulos,
563 E., Fatta-Kassinos, D., 2013a. Winery wastewater purification by reverse osmosis and oxidation
564 of the concentrate by solar photo-Fenton. *Separation and Purification Technology* 118, 659-669.
565 <http://dx.doi.org/10.1016/j.seppur.2013.07.049>

566 Ioannou, L.A., Puma, G.L., Fatta-Kassinos, D., 2015. Treatment of winery wastewater by
567 physicochemical, biological and advanced processes: A review. *Journal of Hazardous Materials*
568 286, 343-368. <https://doi.org/10.1016/j.jhazmat.2014.12.043>

569 ISO 11348-2:1998. Water quality — Determination of the inhibitory effect of water samples on
570 the light emission of *Vibrio fischeri* (Luminescent bacteria test) — Part 2: Method using liquid-
571 dried bacteria

572 Jorge, N., Teixeira, A.R., Guimarães, V., Lucas, M.S., Peres, J.A., 2021b. Treatment of Winery
573 Wastewater with a Combination of Adsorption and Thermocatalytic Processes. *Processes*, 10, 75.
574 <https://doi.org/10.3390/pr10010075>

575 Jorge, N., Teixeira, A.R., Matos, C.C., Lucas, M.S., Peres, J.A., 2021a. Combination of
576 Coagulation– Flocculation–Decantation and Ozonation Processes for Winery Wastewater
577 Treatment. *Int. J. Environ. Res. Public Health*, 18, 8882. <https://doi.org/10.3390/ijerph18168882>

578 Kaira, W.M., Kimpiab, E., Mpofu, A.B. 2022. Anaerobic digestion of primary winery wastewater
579 sludge and evaluation of the character of the digestate as a potential fertilizer. *Biomass Conv.*
580 *Bioref.* <https://doi.org/10.1007/s13399-022-03087-8>

581 Latorre, A., Malmqvist, A., Lacorte, S., Welander, T., Barceló, D., 2007. Evaluation of the
582 treatment efficiencies of paper mill whitewaters in terms of organic composition and toxicity.
583 *Environmental Pollution* 147, 648-655. <https://doi.org/10.1016/j.envpol.2006.09.015>

584 Lauzurique, Y., Espinoza, L.C., Huiliñir, C., García, V., Salazar, R., 2022. Anodic Oxidation of
585 Industrial Winery Wastewater Using Different Anodes. *Water*, 14, 95.
586 <https://doi.org/10.3390/w14010095>

587 Lucas, M.S., Mouta, M., Pirra, A., Peres, J.A., 2009a. Winery wastewater treatment by a
588 combined process: long term aerated storage and Fenton's reagent. *Water Science & Technology*
589 60.4, 1089-1095. <https://doi.org/10.2166/wst.2009.555>

590 Lucas, M.S., Peres, J.A., Lan, Y.B., Puma, G.L., 2009 b. Ozonation kinetics of winery wastewater
591 in a pilot-scale bubble column reactor. *Water Research* 43, 1523-1532.
592 <https://doi.org/10.1016/j.watres.2008.12.036>

593 Luz S., Rivas J., Afonso A., Carvalho F., 2021. Immediate one-step lime precipitation process for
594 the valorization of winery wastewater to agricultural purposes. *Environmental Science and*
595 *Pollution Research* 28, 18382-18391. <https://doi.org/10.1007/s11356-020-11933-3>

596 Marchão, L., Fernandes, J.R., Sampaio, A., Peres, J.A., Tavares, P.B., Lucas, M.S., 2021.
597 Microalgae and immobilized TiO₂/UV-A LEDs as a sustainable alternative for winery
598 wastewater treatment. *Water research*, 203:15, 117464.
599 <https://doi.org/10.1016/j.watres.2021.117464>

600 Mendonça, E., Picado, A., Paixão, S.M., Silva, L., Cunha, M.A., Leitão, S., Moura, I., Cortez, C.,
601 Brito, F., 2009. Ecotoxicity tests in the environmental analysis of wastewater treatment plants:

602 Case study in Portugal. *Journal of Hazardous Materials* 163, 665-670.
603 <https://doi.org/10.1016/j.jhazmat.2008.07.012>

604 Monteagudo, J.M., Durán, A., Corral, J.M., Carnier, A., Frades, J.M., Alonso, M.A., 2012.
605 Ferrioxalate-induced solar photo-Fenton system for the treatment of winery wastewaters.
606 *Chemical Engineering Journal* 181-182, 281-288. <https://doi.org/10.1016/j.cej.2011.11.080>

607 Moreira, C. 2013. Fotossíntese. *Revista de Ciência Elementar*. 1(01):0003. Fundação Calouste
608 Gulbenkian. ISSN 2183-1270

609 Mosse, K.P.M., Patti, A.F., Smernik, R.J., Christen, E.W., Cavagnaro, T.R., 2012.
610 Physicochemical and microbiological effects of long- and short-term winery wastewater
611 application to soils. *Journal of Hazardous Materials*, 201– 202, 219-228.
612 <https://doi.org/10.1016/j.jhazmat.2011.11.071>

613 Mosteo, R., Ormad, P., Mozas, E., Sarasa, J., Ovelleiro, J.L., 2006. Factorial experimental design
614 of winery wastewaters treatment by heterogeneous photo-Fenton process. *Water Research* 40,
615 1561-1568. <https://doi.org/10.1016/j.watres.2006.02.008>

616 Muñoz-Huerta, R.F., Guevara-Gonzalez, R.G., Contreras-Medina, L.M., Torres-Pacheco, I.,
617 Prado-Olivarez, J., Ocampo-Velazquez, R.V, 2013. A Review of Methods for Sensing the
618 Nitrogen Status in Plants: Advantages, Disadvantages and Recent Advances. *Sensors* 2013, 13,
619 10823-10843. <https://doi.org/10.3390/s130810823>

620 Netto, A.T., Campostrini, E., Oliveira, J.G., Bressan-Smith, R.E., 2005. Photosynthetic pigments,
621 nitrogen, chlorophyll a fluorescence and SPAD-502 readings in coffee leaves, *Scientia*
622 *Horticulturae*, 104, 2, 199-209. <https://doi.org/10.1016/j.scienta.2004.08.013>

623 Ngwenya, N., Gaszynski, C., Ikumi, D., 2022. A review of winery wastewater treatment: A focus
624 on UASB biotechnology optimisation and recovery strategies. *Journal of environmental chemical*
625 *engineering*, 10:4, 108174. <https://doi.org/10.1016/j.jece.2022.108172>

626 OECD/OCDE, 2006. OECD GUIDELINES FOR THE TESTING OF CHEMICALS. Test No.
627 201: Freshwater Alga and Cyanobacteria, Growth Inhibition Test.

628 Oliveira, M., Costa, J.M., Fragoso, R., Duarte, E., 2019. Challenges for modern wine production
629 in dry areas: dedicated indicators to preview wastewater flows. *Water Supply* 19.2, 653-661.
630 <https://doi.org/10.2166/ws.2018.171>

- 631 Ormad, M.P., Mosteo, R., Ibarz, C., Ovelleiro, J.L., 2006. Multivariate approach to the photo-
632 Fenton process applied to the degradation of winery wastewaters. *Applied Catalysis B: Environmental* 66, 58-63. <https://doi.org/10.1016/j.apcatb.2006.02.014>
- 634 Palma, P., Alvarenga, P., Palma, V., Matos, C., Fernandes, R.M., Soares, A., Barbosa, I.R., 2009.
635 Evaluation of surface water quality using an ecotoxicological approach: a case study of the
636 Alqueva Reservoir (Portugal). *Environ Sci Pollut Res* (2010) 17, 703–716.
637 <https://doi.org/10.1007/s11356-009-0143-3>
- 638 Palma, P., Fialho, S., Alvarenga, P., Santos, C., Brás, T., Palma, G., Cavaco, C., Gomes, R.,
639 Neves, L.A., 2016. Membranes technology used in water treatment: Chemical, microbiological
640 and ecotoxicological analysis. *Science of The Total Environment* 568, 998-1009.
641 <http://dx.doi.org/10.1016/j.scitotenv.2016.04.208>
- 642 Pérez, J., Domingues, I., Soares, A.M.V.M., Loureiro, S., 2011. Growth rate of
643 *Pseudokirchneriella subcapitata* exposed to herbicides found in surface waters in the Alqueva
644 reservoir (Portugal): a bottom-up approach using binary mixtures. *Ecotoxicology* 20, 1167-75.
645 <https://doi.org/10.1007/s10646-011-0661-x>
- 646 Persoone, G., 1999. THAMNOTOXKIT FTM - Crustacean toxicity screening test for Freshwater.
647 Standard Operational Procedure. Belgium. pp 3-21.
- 648 Persoone, G., Marsalek, B., Blinova, I., Torokne, A., Zarine, D., Manusadzianas, L., 2003. A
649 practical and user-friendly toxicity classification system with microbiotests for natural waters and
650 wastewaters. *Environmental Toxicology* 18, 395–402. <https://doi.org/10.1002/tox.10141>
- 651 Riehm, H., 1958. Die ammoniumlaktatessignäure. *Methods zur bestimmung der leichtlöslichen*
652 *phosphorsäure in karbonatigen böden*. *Agrochimica*, 3: 49-65.
- 653 Rizzo, L., Lofrano, G., Belgiorno, V., 2010. Olive Mill and Winery Wastewaters Pre-Treatment
654 by Coagulation with Chitosan, *Separation Science and Technology* 45:16, 2447-2452.
655 <http://dx.doi.org/10.1080/01496395.2010.487845>
- 656 Sánchez-Sastre, L.F., Alte da Veiga, N.M.S., Ruiz-Potosme, N.M., Carrión-Prieto, P., Marcos-
657 Robles, J.L., Navas-Gracia, L.M., Martín-Ramos, P., 2020. Assessment of RGB Vegetation
658 Indices to Estimate Chlorophyll Content in Sugar Beet Leaves in the Final Cultivation
659 Stage. *AgriEngineering*, 2, 128-149. <https://doi.org/10.3390/agriengineering2010009>
- 660 Solís, R.R., Rivas, F.J., Ferreira, L.C., Pirra, A., Peres, J.A., 2017. Integrate aerobic biological-
661 chemical treatment of winery wastewater diluted with urban wastewater. LED-based

662 photocatalysis in the presence of monoperoxysulfate. *Journal of Environmental Science and*
663 *Health, Part A*, 53:2, 124-131. <https://doi.org/10.1080/10934529.2017.1377584>

664 Sousa, R.M.O.F., Amaral, C., Fernandes, J.M.C., Fraga, I., Semitela, S., Braga, F., Coimbra, A.M.,
665 Dias, A.A., Bezerra, R.M., Sampaio, A., 2019. Hazardous impact of vinasse from distilled
666 winemaking by-products in terrestrial plants and aquatic organisms. *Ecotoxicology and*
667 *Environmental Safety* 183, 109493. <https://doi.org/10.1016/j.ecoenv.2019.109493>

668 Ünlükara, A., Cemek, B., Karaman, S., Erşahin, S., 2008. Response of lettuce (*Lactuca sativa* var.
669 *crispa*) to salinity of irrigation water, *New Zealand Journal of Crop and Horticultural Science*,
670 36:4, 265-273. <https://doi.org/10.1080/01140670809510243>

671 Yang, L., Ying, G., Su, H., Stauber, J., Adams, M.S., Binet, M.T., 2008. Growth-inhibiting effects
672 of 12 antibacterial agents and their mixtures on the freshwater microalga *Pseudokirchneriella*
673 *subcapitata*. *Environmental Toxicology* 27, 1201-1208. <https://doi.org/10.1897/07-471.1>

674 Zar, J.H., 1996. *Biostatistical analysis*. USA, DC: Prentice-Hall International, Englewood Cliff.

675 Zgórska, A., Arendarczyk, A., Grabińska-Sota, E., 2011. Toxicity assessment of hospital
676 wastewater by the use of a biotest battery. *Archives of Environmental Protection* 37, 55-61.
677 <https://doi.org/10.12912/23920629/112650>

678