

Improving crunchiness and crispness of fried squid rings through innovative tempura coatings: addition of alcohol and CO₂ incubation

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Abstract This study aimed to determine the effect of different uncommon tempura formulations (incubated with CO₂ and with added ethanol) on physical, chemical and sensory characteristics of fried coated squids, immediately after frying and also after 48 h of refrigeration storage and subsequent oven reheating. Ethanol addition led to lower levels of moisture and higher of fat in the fried coating, regardless the use of CO₂. There were no difference in instrumental colour parameters among all four battered samples. Ethanol added tempuras showed higher crispness and lower maximum force than their water counterparts in an instrumental texture evaluation. Fried squids coated with ethanol added tempuras were rated as crispier and crunchier, even after 2 days of storage and further reheating. Overall, partial substitution of water by ethanol appears as an interesting strategy to increase crispness in tempura coated fried products, especially when the products are stored and reheated before consumption.

Keywords Deep fat-frying · Oven reheating · Tempura · Crispness · Ethanol

Introduction

Deep fried coated foodstuffs are known and appreciated worldwide (Fizman and Salvador 2003) constituting some of the most extensively consumed convenience food products (Varela et al. 2008). Coating preserves and enhances food quality, limiting moisture lost during frying and contributing to the generation of pleasant flavors (Firdevs Dogan et al. 2005; Varela and Fizman 2011). The crusts of fried products exhibit a characteristic crunchy texture and a golden yellow color (Chen et al. 2009). A crunchy and crispy crust with a tender and moist inside, are to a great extent, responsible for their great acceptance (Fizman and Salvador 2003).

Crispness is a highly valued textural characteristic in breaded and battered fried foods, such as fish, poultry or vegetables (Firdevs Dogan et al. 2005). In fact, it has been reported as the most critical property determining their consumer acceptance (Maskat and Kerr 2002; Primo-Martín and van Deventer 2011). Consequently, extensive research has been conducted to investigate the influence of batter ingredients on crispness (Fizman and Salvador 2003). In fact, tempura-type batter mixtures widely vary depending on the type of food being batter-coated and the specific characteristics desired for the final products (Salvador et al. 2005). Numerous industrial and culinary attempts to produce crispy and crunchy tempura crusts have been assayed.

A batter can be defined as a liquid mixture composed of water, flour, starch, salt, leavening and other minor ingredients, into which food products are dipped prior to deep-frying (Fizman 2009). Flour functionality in batter systems largely depends upon the two major constituents of all flours: starch and protein. The proteins in batter provide structure and increase the coating pick-up values and final

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yield in fried products (Fizman and Salvador 2003). The main protein in wheat flour responsible for such functionality is gluten. Gluten is a tough, elastic substance that acts as a net, trapping and holding air bubbles in the batter, and contributing to the coherence of the batter and to the adhesion to the battered products (Mallikarjunan et al. 2009). Depending on the quantity, quality and extent of hydration of gluten, the resultant structure can range from a crispy and brittle structure as in bread crust, to an elastic one, as it happens in bread loafs, or even a flowing one, as in a tempura, thus determining the products final texture (Schiffmann 2011).

Tempura coated fried foods are crispy right after frying, due to the formation of a mostly dehydrated gluten network layer. As moisture increases, it plasticizes the structural matrix, eventually leading to an elastic structure that results soggy and not crispy (Schiffmann 2011). Such moisture may come from the surrounding atmosphere during storage or holding after frying, or from the coated food, during reheating in the oven or in the microwave.

There have been numerous attempts to keep crispness in coated fried foods, by modifying the formula, using hydrocolloids or different types of flours (Fizman 2009), with very limited success, especially when it comes to reheating. Blumenthal (2006) developed a tempura, which principle is based on the substitution of part of the water in the batter with ethanol. More specifically, he suggested mixing wheat flour with vodka and beer. Moreover, in such a recipe it was indicated that the tempura should be infused with CO₂ into a siphon, in order to achieve more bubbles in the batter, that will eventually lead to more layers in the coating of the fried product. The author also claimed that during storage and further reheating, the fried tempura remains crispy.

Thus, the objective of this study was to shed some light on the effect of modifying tempuras by partially substituting water with ethanol and by infusing the liquid tempura with CO₂, on several physical, chemical and sensory characteristics of tempura fried calamari, right after the deep-frying process and after storage and further reheating in the oven, with special attention to crispness.

Materials and methods

Experimental design

Frozen squid rings were purchased in a local supermarket and were thawed before coated with four different tempura formulations. The tempuras were produced using a regular wheat flour for frying. The four different formulations corresponded to the use of (1) either only mineral water (groups named as “H₂O”) or a mixture of water with

ethanol (named as “Eth”) for food use (National Distillery, Riachos, Portugal) and (2) to the incubation or not of the tempuras in a siphon charged with CO₂ (Isi- Consumer Products, Vienna, Austria), (groups named respectively as either “no-CO₂” or “CO₂”).

Tempuras were produced by manually mixing 250 g of flour with 375 mL of either water or a water/ethanol mixture (225 mL water/150 mL ethanol) with a spoon for 2 min. Right after, those tempuras subjected to CO₂ incubation were poured into a siphon and incubated with 4 charges of CO₂ for 180 min. Thereafter, the content was siphoned onto a bowl. Tempuras non containing CO₂ were also left to rest for the same time.

Around 60 squid rings were dipped into each type of tempura for 4–6 s. Battered squid rings were deep-fried on sunflower oil at 190 ± 5 °C during 210 s in a domestic fryer (Philips, Germany). Half of the fried tempura squids from each batch were kept in a tray covered with transparent film at 4 °C for 48 h. After this time, the four different battered type of samples were reheated at 130 ± 5 °C during 15 min in a conventional domestic convection oven (UFESA, Maxi-Grill BSHTAESL). The full experiment was replicated three times.

Sensory analysis of the full fried tempura squid (crust and squid ring) immediately after frying or oven reheating. In 30–40 samples, coating and squid were separated in order to proceed to analyse the coating (3 squid rings gave approx 10 g of coating). In both cases, temperature of coating or battered squid rings were continuously controlled with the intention of never being less than 45 ± 5 °C.

Moisture content

Moisture content of the coating was determined by drying the samples (aprox. 1 g) at 102 °C into an oven (AOAC 2000). Five replicates were performed for each kind of formulated batter.

Fat content

The total oil content of the crusts was determined by extraction with petroleum ether using a Soxtec Avanti 2050 extraction system (Foss Tecator, Sweden). Previous hydrolysis of samples was carried out with HCl and fat was calculated as a percentage. Five replicates per sample were used.

Color parameters

Color measurements were made on each of four randomly selected battered squid rings from each formulation for each storage time (0 and 2 days). The Commission

Internationale d'Eclairage (CIE) L^* , a^* and b^* values were determined using a portable spectrophotometer (Konica Minolta CM-600d, Osaka, Japan) that was calibrated with a standard white calibration tile. Fifteen replicates were performed for each batch of samples.

Instrumental texture analysis

A TA-HDi Texture Analyser (Stable Micro Systems, Godalming, UK) equipped with the Kramer Shear Cell with 5 blades (HDP/KS5), was used for the compression tests with a 25 kg load cell. Kramer analysis was made just after the cooking process (deep-frying or oven convection reheating), in order to assure crust temperature to be around 45 ± 5 °C.

The following settings were used for the analysis: test speed 2 mm/s, trigger force 0.05 N, travel distance of the blades 15 mm and acquisition points per second 400 pps. Ten replications with approximately 10 g of fried tempura were performed for each type.

The following parameters were calculated from the force versus time curves: maximum force (N) respective work force or area (Ns), number of peaks above in the curve or fracture events (drop in force higher than 0.049 N), and initial slope (N/s) or gradient (slope of the curve up to the first major peak) (Sanz et al. 2007).

Scanning electron microscopy (SEM)

Samples for scanning electron microscope (SEM) observations were first washed with ether diethyl and then dried with a nitrogen stream, and afterwards were sectioned and the surface was coated with gold using a gold coater under vacuum, before taking the micrographs. SEM Quanta 3D FEG (FEI Company) micrographs were taken with a secondary electrons (SE) detector, at an accelerating voltage of 2 or 5 kV, a working distance varying from 10 to 11.4 mm and an average magnification of 150–5000 times.

Sensory analysis

A panel of 15 assessors with experience in the descriptive evaluation of crispy products was used to evaluate the four samples of battered squid rings at production day and with 48 h at reheated oven samples. Panellists were trained for 6 months in group sessions; in a first stage, group discussion with a variety of samples showing a wide range of crispness and crunchiness were carried out, in order to achieve a consensus on the meaning and intensity of these attributes (hardness, crispness, crunchiness). Thereafter, assessors were trained in sessions in individual booths in which the repeatability was addressed.

Testing was carried out in a sensory laboratory equipped with individual booths (NP ISO 8586, 2012) in different sessions in function of the production day (day 0 or reheated after 48 h). Surface temperature of the samples was controlled by using an infrared thermometer (Testo, Germany) in order to ensure 45 ± 5 °C. Four samples were given to each panellist in each session, corresponding to the four formulas of batters evaluated, either right after frying or after 48 h of refrigeration storage and further regeneration in the oven. Samples were given in random order on plastic plates and identified with a random letter code. Panellists were instructed to rinse their mouths with water between sample evaluations.

The intensities of sensory attributes were scored on 9 cm unstructured line scales labelled from “low” (0) to “high” (9). Color intensity and its homogeneity were considered as the color ranging from yellow to gold to brown and how uniform is the surface of the coating of the fried tempura squid after the thermal process (Albert et al. 2009). Adherence of tempura to squid was visually evaluated along the crust perimeter of the cutting surface after biting (Albert et al. 2009). To evaluate hardness, the instruction was to bite the whole sample with the incisor teeth until fracture and score the material resistance (Vincent 2004). To score crunchiness the instruction was to evaluate the intensity and pitch of the sound produced during mastication (first three bites) (Dijksterhuis et al. 2007). To evaluate crispness, the instruction was to score the number of layers with incorporated air during chewing (Dijksterhuis et al. 2007). Crumbliness was evaluated as the degree to which the sample fractures into pieces (Pascua et al. 2013). Oiliness was assessed as the amount of fat and oil in the mouth during chewing (Albert et al. 2009). Flavour intensity was evaluated as the strength of flavour perception during chewing and right after swallowing.

Statistics

The effects of partial substitution of water with ethanol and of either incubate or not the tempura with CO₂ on chemical, instrumental colour and instrumental texture parameters were analysed by a two-way analysis of variance (ANOVA) using the GLM procedure of SPSS 12.0.1 (SPSS Inc., Chicago, Illinois, US). Differences between mean values were evaluated at the 0.05 level using the Tukey pair-wise comparison test. In order to evaluate the effect of refrigeration and subsequent oven reheating, a one-way ANOVA considering time as the only factor within each type of fried tempura squids, was also performed.

For the data reported by the panellists, a mixed analysis of variance was used, with ethanol addition and CO₂ or storage time as fixed effects, plus the random factors of tasting session and panellist.

Results and discussion

Moisture and fat content

Moisture content on freshly fried tempura samples (day 0) and on oven reheated samples after 48 h refrigeration (day 2), in samples processed with four different formulations (either only with water or partially substituted with ethanol, and either or not incubated with CO₂) are shown on Table 1.

Substitution of water with ethanol, CO₂ incorporation and their interaction significantly affected moisture content of the fried coated tempura. Samples produced with just water showed lower moisture content when incubated with CO₂ (43.4 and 38.9%, no-CO₂ and CO₂, respectively), while those elaborated with ethanol showed higher levels due to CO₂ (11.6 and 18.4%, no-CO₂ and CO₂, respectively). The same trend was shown for reheated samples after 2 days of refrigerated storage, with fried tempura samples made just with water showing higher moisture contents when not incubated with CO₂ (40.9 and 34.7%, no-CO₂ and CO₂, respectively), while those made with added ethanol showed no effect due to CO₂ (15.1 and 14.9%, no-CO₂ and CO₂, respectively). CO₂ seemed to differently affect moisture content of tempura made just with water or with added ethanol, both, right after frying and after 2 days of storage and further reheating. We find no clear explanation for that. To our knowledge, this is the first study in which these two factors are considered, and there is no previous scientific literature on the matter. A structure with more bubbles due to the presence of CO₂ could ease dehydration during frying, but if this were the case, it would be difficult to understand why it did not behave in the same way in the ethanol added tempura on day 0.

Formulations with added ethanol, with or without CO₂, showed lower moisture content than those just containing

water. This is most likely due the far lower water content of the original batter before frying in the tempuras that had added ethanol. Water loss in both types of tempuras during frying was similar: the initial calculated water content of both batters were 60 and 36% (for H₂O and Eth tempuras, respectively), while the content after frying was between 43.4 and 38.9% in H₂O tempuras, and between 11.6 and 18.4% in Eth ones. This means that water loss during frying was around 20% for both types.

All four type of samples had significant differences from day 0 to day 2 in moisture content, but with an opposite behaviour in H₂O and Eth ones: while in H₂O ones there was a decrease in water content (both in CO₂ and no-CO₂ ones), fried tempuras with added ethanol showed a decrease in moisture due to storage and further reheating only in CO₂ samples, while no-CO₂ ones, that were those showing the lowest moisture content, showed an increase. During refrigeration, the water content equilibrates between the food and the surrounding atmosphere (Mercier et al. 2015), which would explain why samples with higher water content tended to lose moisture, while those with a lower water content showed an increase.

On top of that, water transfer between different ingredients or parts of the food unit also takes place if their water activities are different (fried batters and squid core). Depending on the temperature and relative humidity at which the product is exposed, the final moisture content at equilibrium will be defined by the sorption isotherms, and thus the kinetics of equilibration will again depend on water diffusion rate (Mekprayoon and Tangduangdee 2012). In the case of those tempuras containing just water this seemed to be not enough to counteract the evaporative loss during refrigeration and heating, while in Eth/No-CO₂ ones could have contributed to the gain in water content.

As far as fat content is concerned (Table 1) addition of ethanol to the tempura significantly ($p \ll 0.001$) increased fat content of the fried tempura, both at day 0 and after

Table 1 Moisture and fat content (%) on fried tempura squids right after frying (day 0) and on samples reheated in the oven after 2 days of storage (day 2), processed with the four studied formulations: with just water or with added ethanol, and with or without CO₂

	H ₂ O		Eth		SEM	p_{eth}	p_{CO_2}	$p_{\text{eth} \times \text{CO}_2}$
	No-CO ₂	CO ₂	No-CO ₂	CO ₂				
Moisture								
Day 0	43.4 ^a	38.9 ^b	11.6 ^d	18.4 ^c	3.1	0.001	0.026	0.001
Day 2	40.9 ^a	34.7 ^b	15.1 ^c	14.9 ^c	2.6	0.001	0.001	0.001
p_{time}	0.001	0.001	0.001	0.001				
Fat								
Day 0	12.9 ^b	18.1 ^b	31.7 ^a	32.0 ^a	2.1	0.001	0.151	0.159
Day 2	16.3 ^d	25.8 ^c	29.4 ^b	36.3 ^a	1.6	0.001	0.001	0.011
p_{time}	0.150	0.001	0.010	0.001				

Cells in the same row with different superscript are significantly different ($p \ll 0.05$) in the Tukey's test
SEM standard error of the mean

48 h of refrigeration storage and reheating. However, in this second case, there was a significant interaction ($p < 0.001$) between ethanol addition and CO₂. At any rate, those tempuras containing ethanol showed far higher fat content right after frying (12.9–18.1% for H₂O ones and 31.7–32.0% for Eth ones, with and without CO₂ respectively), and after storage and reheating (16.3 and 25.8% for H₂O ones and 29.4–36.3% for Eth ones, with and without CO₂ respectively). Overall, these effects were in agreement with moisture content results, so that a decrease or increase

in water content was paralleled with an increase or decrease, respectively, in fat content. But on top of this, there might be an effect of ethanol on increasing the oil uptake. After immersion into hot oil, the temperature of the surface layers rises rapidly. Water starts boiling at 100 °C and it is released from the surface as steam bubbles. As frying proceeds, the thickness of the formed crust continues to increase. It seems that during the cooling phase, the superheated steam in the bubbles on the surface of the product cools down, leading to a pressure drop that sucks the oil into the void spaces (Achir et al. 2008). In the case of ethanol containing tempura samples, the faster evaporation of ethanol and its higher diffusivity would have probably led to a higher number of empty voids in the surface, which could eventually lead to a higher oil uptake. Such an increasing effect of ethanol containing tempura on the final fat content of fried battered foodstuff poses clear consequences on the nutritional outcome of the product, leading to a higher caloric value, which might be not adequate in those consumers aiming to reduce such. However, given that fried battered food are intrinsically of high caloric content, this might be not relevant.

The increase or decrease in fat content from day 0 to day 2 was most likely due to the variations in the moisture content, so that if there was a decrease in water, there occurred a proportional increase in the other components.

Textural analysis

Force versus time parameters obtained after crushing fried tempura samples in a Kramer cell are shown on Table 2. All studied parameters (maximum force, work load and number of fracture events) showed a significant interaction between ethanol addition and CO₂, both on day 0 and on day 2, except for the maximum force on day 2 and the number of breaking events on day 0.

H₂O/no-CO₂ samples showed higher maximum force values than all the other groups, both on day 0 and on day 2. H₂O/CO₂ fried tempura samples had lower values on work load than any of the other groups on day 2, while on day 0 such a difference was only significant with the H₂O/no-CO₂ samples. Neither maximum force, nor work load,

showed any significant variation after 48 h storage and oven reheating.

Eth samples showed significant higher number of fracture events than H₂O ones right after frying and also after 48 h storage and further re-heating in the oven. Reheated H₂O/CO₂ samples showed lower number of breaking events than their H₂O/no-CO₂ counterparts, but CO₂ showed no effect on Eth samples on any of the day tested. Only H₂O/CO₂ samples showed a significant reduction in the number of breaking events from day 0 to day 2, while the other groups experimented no variation in this parameter.

Mechanical properties obtained after compression tests allow inferring the structural properties of the product and how they will be perceived during chewing (Saeleaw and Schleining 2011). Compression tests using different probes and parameters have shown good correlations with crispness and crunchiness for some products, but not for others. Compression in a Kramer cell has been extensively used for measuring crispness in different types of dry snacks (Chaunier et al. 2005; Sandoval et al. 2008). Some authors have shown that both, maximum force and work load to failure, show an inverse correlation with sensory crispness (Van Hecke et al. 1998). Usually, a brittle product will also exhibit numerous fracture events, followed by sudden drops in force as the crack propagates (Miranda and Aguilera 2006), so that the higher the number of fracture events, the higher the crispness (Saeleaw and Schleining 2011).

In our samples, the use of ethanol in the tempura led to a clear increase in the indicators for crispness: lower maximum force and work load, and higher number of fracture events. This last fact is notable, since the average number of events outnumbered the regular fried tempura in more than 100%. And even more interesting, such indicators remained quite stable after 2 days of storage and further reheating. This is remarkable, since one of the main problems with storage and reheated fried coated foodstuff is the loss of crispness. The retention of crispness for some time after frying is an important quality factor in fried battered products (Baixauli et al. 2003). We could objectively detect the effect of ethanol on fried tempura crispness claimed by Blumenthal (2006). Nevertheless, it could be seen that there was not any significant modification in the indicators for crispness in samples not containing ethanol, with the exception of the significant decrease in the number of breaking events of H₂O/CO₂ fried tempuras.

A potential explanation for the effect of ethanol on the mechanical indicators of crispness could be the fact that, during frying, alcohol evaporation is faster and more violent than that of water, which might lead to the formation of a higher number of bubbles. Usually, the number of bubbles has been positively linked to crispness (Miranda

Table 2 Parameters from the force displacement curves obtained by compression on fried tempura right after frying (day 0) and on samples reheated in the oven after 2 days of storage (day 2), processed with the four studied formulations: with just water or with added ethanol, and with or without CO₂

	H ₂ O		Eth		SEM	<i>p</i> _{eth}	<i>p</i> _{CO₂}	<i>p</i> _{eth × CO₂}
	No-CO ₂	CO ₂	No-CO ₂	CO ₂				
Max. force (N)								
Day 0	2.0 ^a	1.5 ^b	1.4 ^b	1.4 ^b	0.06	0.020	0.041	0.022
Day 2	1.9 ^a	1.4 ^b	1.4 ^b	1.3 ^b	0.06	0.001	0.001	0.081
<i>P</i> _{time}	0.827	0.394	0.885	0.108				
Work load (N s)								
Day 0	10.0 ^a	7.3 ^b	8.5 ^{ab}	8.6 ^{ab}	0.33	0.898	0.035	0.048
Day 2	10.8 ^a	6.7 ^b	10.5 ^a	9.4 ^a	0.37	0.031	0.001	0.005
<i>P</i> _{time}	0.269	0.311	0.110	0.326				
No. of fracture events								
Day 0	100.2 ^b	114.4 ^b	243.3 ^a	220.4 ^a	11.53	0.001	0.877	0.392
Day 2	80.6 ^b	44.1 ^c	217.2 ^a	247.3 ^a	14.11	0.001	0.707	0.001
<i>P</i> _{time}	0.148	0.001	0.524	0.196				

Cells in the same row with different superscript are significantly different ($p \leq 0.05$) in the Tukey's test
SEM standard error of the mean

and Aguilera 2006). However, in our study, those samples containing CO₂, that are supposed to have a larger number of bubbles, did not show a clear tendency to show better indicators for crispness. Therefore, it seems that the increased crispness in ethanol added tempura is not only related to the larger number of bubbles as a consequence of sudden evaporation of ethanol during frying. Actually, crispness in snacks and other dry crispy products is very much related to the extent of dehydration (Katz and Labuza 1981). In our case, the Eth samples got much more dehydrated than H₂O ones, and this could also contribute to their higher values for crispness indicators. Moreover, in the specific case of wheat, such a lower amount of available moisture could help to reach the glass state of the dried gluten network (Kokini et al. 1994), and this effect could last in time even after storage and reheating. During this processes, the water from the core (squids in this study) plasticizes the gluten network, reaching again the rubbery state, which eventually leads to an elastic structure and loss of crispness, as it happens during the staling of bread (Primo-Martín and van Deventer 2011). The use of ethanol seems to prevent such a process, either due to the very low levels of moisture achieved in the fried tempura, or to the lack of initial moisture to reach a complete development of the gluten network.

SEM

Images taken using SEM are shown in Fig. 1. Both ethanol addition and CO₂ led to a less compact structure, with more bubbles and tiny holes. Also, ethanol samples showed a smoother surface regardless the presence of CO₂. Such a structure agrees with the obtained results on the

instrumental texture analysis tests, showing a more aerated structure as a result of the sudden evaporation of ethanol and the presence of CO₂ bubbles.

Pores development is one of the main structural changes during deep-fat frying. Evaporation of moisture from the tempura due to the high temperature during frying creates capillary paths on its way out of the product. These can even be large cavities and crevasses due to explosive evaporation, especially if there is quick formation of crust (Ngadi et al. 2008). This is actually what seemed to happen in our samples, since the ethanol added ones showed the sign of larger cavities and crevasses, probably as a result of (1) the sudden and more violent evaporation of ethanol and (2) the faster formation of the crust, as a result of the lower amount of water that needed to be evaporated to reach the glass state.

Color parameters

Colour parameters L*, a*, b* for fried tempura samples processed with the four different studied formulations (either with just water or with added ethanol, and with or without CO₂) on day 0 and after 2 days of storage and further reheating in the oven (day 2), are shown in Table 3.

The interaction of ethanol and CO₂ significantly influenced L* values of fried tempura samples on day 0 and after storage and reheating of the samples ($p = 0.021$ and $p = 0.018$, respectively). Overall, H₂O samples showed higher L* values than their ethanol added counterparts, especially in those without CO₂. However, CO₂ did not show any effect on the L* values of samples right after frying, and the opposite one on samples stored and reheated.

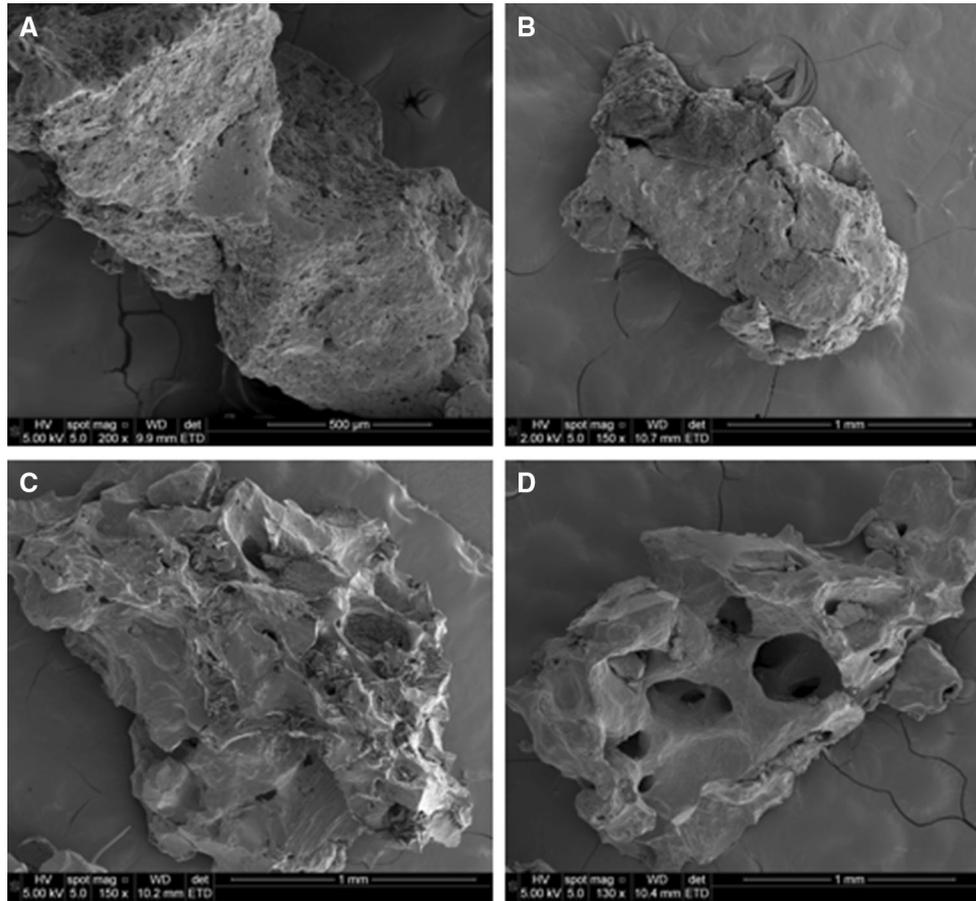


Fig. 1 SEM images of different types of fried tempuras: (a H₂O/no-CO₂; b H₂O/CO₂; c eth/no-CO₂; d eth/CO₂)

Table 3 Instrumental colour parameters (L*, a* and b*) on fried tempuras right after frying (day 0) and on samples reheated in the oven after 2 days of storage (day 2), processed with water or with added ethanol, and with or without CO₂

	H ₂ O		Eth		SEM	P _{eth}	P _{CO₂}	P _{eth & CO₂}
	No-CO ₂	CO ₂	No-CO ₂	CO ₂				
L*								
Day 0	66.9 ^a	57.3 ^b	57.8 ^b	57.1 ^b	1.02	0.018	0.007	0.021
Day 2	60.8 ^a	57.1 ^{ab}	52.4 ^b	55.2 ^b	0.90	0.001	0.685	0.018
P _{time}	0.136	0.789	0.019	0.147				
a*								
Day 0	- 2.8 ^a	- 1.8 ^{ab}	- 0.4 ^c	- 0.8 ^{bc}	0.25	0.004	0.613	0.435
Day 2	- 1.2 ^b	- 1.0 ^b	1.0 ^a	- 0.2 ^{ab}	0.27	0.001	0.207	0.067
P _{time}	0.074	0.185	0.061	0.418				
b*								
Day 0	22.7 ^{ab}	23.5 ^a	20.3 ^{ab}	19.3 ^b	0.58	0.007	0.468	0.603
Day 2	21.6 ^{ab}	23.7 ^a	18.0 ^b	19.7 ^{ab}	0.67	0.002	0.076	0.904
P _{time}	0.545	0.952	0.206	0.923				

Cells in the same row with different superscript are significantly different ($p < 0.05$) in the Tukey's test SEM standard error of the mean

As compared to other studies on tempura fried squids (Salvador et al. 2005; Sanz et al. 2004), the L* values were similar, although different factors not directly related to

those evaluated in the present study could lead to small differences (type of oil, ingredients in the tempura and so on). As far as the effect of storage and reheating is

concerned, only Eth/no-CO₂ samples showed a significant decrease in L* values ($p = 0.019$), while the other types remained unaffected. So, overall, it seems that the presence of ethanol led to a darker surface. Within H₂O tempuras, CO₂ also led to darker colours on day 0, but not on day 2. In the case of ethanol, such an effect could be related to a lower water activity, which in turn could have positively influenced the development of browning reactions (van Boekel 2001). However, such a hypothesis does not support the effect of CO₂ on lightness. It could be that the higher number of bubbles on the crust may lead to a different reflection of the light, which could also explain the lower lightness values on ethanol added samples. Changes during the storage and reheating (dehydration and further browning reactions) seemed to affect only samples with added ethanol and without CO₂.

Colour parameters a* and b* showed a significant effect of ethanol addition, both in day 0 and in day 2 ($p = 0.004$ and $p = 0.001$ for a* on days 0 and 2, respectively; $p = 0.007$ and $p = 0.002$ for b* on days 0 and 2, respectively). Neither CO₂, nor its interaction with Eth showed a significant effect on these two parameters.

It was expected that a* and b* values were higher in samples at day 2, since undergoing two thermal processes (frying and oven) most likely to a stronger development of browning reactions than just a single one (frying). Also, dehydration of the surface may have also caused higher a* and b* values after storage and reheating. However, these processes seem to be not enough to make any difference between freshly fried and reheated products.

Sensory analysis

Results from the sensory analysis of fried squid rings battered with either a water or ethanol based tempura, incubated or not with CO₂, and freshly fried or reheated after 2 days of refrigeration storage, are shown in Table 4. Ethanol addition and CO₂ showed a significant interaction on colour homogeneity, crispness, crunchiness, crumbliness and oiliness on day 0, but not on day 2. As individual factors, CO₂ did not affect any of the sensory parameters assessed, while the addition of ethanol significantly modified adherence of the tempura to the squid on day 2, hardness on both days, crispness, crunchiness and crumbliness on day 2 and flavour intensity on both days. Storage and further reheating significantly decreased the adherence of the tempura to the squids on H₂O/no-CO₂ samples, the crunchiness on ethanol added tempura samples and the crispness and crumbliness in both types of CO₂ incubated samples. On the other hand, oiliness was lower in H₂O/no-CO₂ samples as a result of storage and reheating.

As long as the external colour of the samples is concerned, sensory results only partially confirmed instrumental colour results: samples with added ethanol tended to have higher colour intensities on day 0, but not on day 2. The mentioned positive effect of lower water activity values on browning could be the reason for this effect. Nevertheless, such an effect is not dramatic in terms of scores. The homogeneity of the colour showed an interaction effect, so that samples without ethanol and with CO₂ showed the highest values, while Eth/CO₂ showed the lowest.

There was a better adherence of the coating to the squid in the ethanol based tempuras on day 2, while this effect was not detected on samples right after frying. It could be that the lower elasticity of the crust in ethanol added samples might have led to a more stable interaction between the coating and the squid, while the plasticiser effect of water in the non-ethanol ones, leading to a more elastic coating, might have eased their separation.

Hardness scores of samples with ethanol added tempura were clearly higher than their water counterparts on both days, while CO₂ did not show a clear effect. This evidences that the perception of hardness in products that are intrinsically crunchy is not related to the maximum instrumental force, since the latter parameter can actually be very high as a consequence of a highly elastic structure. In fact, as commented before, maximum force and work load have been previously related to lower values of crunchiness. This was confirmed with the evaluation of crispness and crunchiness of tempura fried squids: ethanol addition boosted the scores of both parameters, confirming the results obtained for the instrumental texture evaluation. The fact that both crispness and crunchiness resulted higher in ethanol based fried tempuras points out to the occurrence of the previously discussed phenomena in samples containing ethanol: violent evaporation leading to a higher number of bubbles and formation of a stable glassy state due to lower moisture content (Ngadi et al. 2008; Primo-Martín and van Deventer 2011). CO₂ incubation, which is supposed to lead to a higher number of bubbles, only have a significant effect on the crispness of freshly fried squids on water based tempuras, while it showed no effect on day 2, nor on ethanol ones in any of the days. It might be that the potential effect of CO₂ on crispness is masked due to the overwhelming effect of ethanol on the same parameter. On the water based samples, its effect might be dependent upon the presence of a glassy structure in the walls of the bubbles: when such a structure becomes rubbery, the crispness is impaired despite of the number of bubbles. Accordingly, H₂O/CO₂ samples were crispier than their no-CO₂ counterparts on day 0, but not after 2 days of storage and further reheating.

Table 4 Average scores for the evaluated sensory parameters of the fried tempura squids right after frying (day 0) and on samples reheated in the oven after 2 days of storage (day 2), processed with the four studied formulations: with just water or with added ethanol, and with or without CO₂

	H ₂ O		Eth		SEM	<i>p</i> _{eth}	<i>p</i> _{CO₂}	<i>p</i> _{eth × CO₂}
	No-CO ₂	CO ₂	No-CO ₂	CO ₂				
Colour intensity								
Day 0	4.1	3.7	5.2	5.2	0.3	0.051	0.727	0.694
Day 2	4.5	4.8	5.7	5.0	0.2	0.142	0.703	0.293
<i>p</i> _{time}	0.663	0.232	0.454	0.691				
Colour homogeneity								
Day 0	5.5 ^{ab}	6.3 ^a	5.2 ^{ab}	4.8 ^b	0.2	0.008	0.451	0.035
Day 2	5.8	5.5	4.8	5.1	0.2	0.109	0.977	0.446
<i>p</i> _{time}	0.597	0.154	0.333	0.384				
Tempura adherence								
Day 0	4.8	4.3	4.9	4.6	0.1	0.574	0.187	0.627
Day 2	3.7 ^b	4.3 ^{ab}	4.9 ^a	5.0 ^a	0.2	0.001	0.068	0.287
<i>p</i> _{time}	0.007	0.982	0.986	0.178				
Hardness								
Day 0	3.4 ^b	3.9 ^b	5.4 ^a	5.3 ^a	0.3	0.001	0.461	0.219
Day 2	3.8 ^b	4.3 ^{ab}	5.4 ^a	5.4 ^a	0.3	0.002	0.405	0.390
<i>p</i> _{time}	0.563	0.381	0.987	0.414				
Crunchiness								
Day 0	3.1 ^b	4.1 ^b	7.1 ^a	6.8 ^a	0.5	0.001	0.181	0.039
Day 2	2.7 ^b	3.0 ^b	5.8 ^a	5.4 ^a	0.4	0.001	0.805	0.276
<i>p</i> _{time}	0.401	0.062	0.049	0.005				
Crispness								
Day 0	3.0 ^c	4.6 ^b	6.7 ^a	6.8 ^a	0.5	0.001	0.007	0.019
Day 2	2.8 ^b	3.0 ^b	5.6 ^a	5.2 ^a	0.4	0.001	0.812	0.374
<i>p</i> _{time}	0.667	0.025	0.098	0.003				
Crumbliness								
Day 0	2.7 ^c	3.6 ^b	5.8 ^a	6.0 ^a	0.43	0.001	0.007	0.029
Day 2	2.5 ^b	2.8 ^b	4.8 ^a	4.6 ^a	0.33	0.001	0.891	0.414
<i>p</i> _{time}	0.587	0.024	0.067	0.009				
Oiliness								
Day 0	4.5 ^b	5.8 ^a	5.7 ^a	6.0 ^a	0.19	0.001	0.001	0.008
Day 2	5.1	5.7	5.5	5.7	0.11	0.242	0.094	0.339
<i>p</i> _{time}	0.012	0.692	0.645	0.322				
Flavour intensity								
Day 0	3.3 ^b	3.8 ^{ab}	5.7 ^a	5.7 ^a	0.39	0.002	0.661	0.622
Day 2	3.6 ^b	4.1 ^{ab}	5.5 ^a	5.2 ^a	0.27	0.002	0.675	0.270
<i>p</i> _{time}	0.736	0.743	0.482	0.275				

Cells in the same row with different superscript are significantly different ($p \leq 0.05$) in the Tukey's test
SEM standard error of the mean

Crumbliness showed a similar trend to that of crispness. The higher crumbliness of ethanol based fried tempuras might be related with the vitreous state attained during frying, due to a more intense dehydration: the vitreous structure would be more difficultly to get rehydrated by saliva while chewing, so that the pieces of crust formed during mastication will be larger, glassier, will produce a louder noise upon breaking and will be not so easy to swallow. On the other hand, the plasticiser effect of water

and saliva would have made the water based samples soggy during chewing,

After storage and reheating, ethanol based tempuras showed, or tended to show, lower values of crispness, crunchiness and crumbliness, but they were still showing much higher values than water based ones.

Both ethanol types (with and without CO₂) and H₂O/CO₂ showed higher scores for oiliness than the plain traditional tempura samples on day 0. On day 2 the trend was

similar, but not significant. Indeed, ethanol based tempura samples were those presenting higher fat contents (Table 1), while within water based ones, CO₂ incorporation also led to an increase in fat content. Therefore, it seems that changes in oiliness perception were mostly due to a different oil content in tempuras, as a result of a higher absorption of oil due to a more porous structure and to a lower moisture content.

Fried tempura squids with ethanol showed higher flavour intensity scores than water based ones. This could be somehow related to the more intense development of Maillard reactions during frying due to lower water contents in these samples, as it was pointed out for colour intensity. However other factors should not be dismissed: ethanol based samples absorbed more oil, which itself is a rich source of flavour compounds in deep-fat fried products (Warner 2008). Moreover, the presence of a higher amount of fat can also affect the release of volatile flavour compounds (Guichard 2002), modifying the profile of compounds reaching the sensitive epithelium in the nose.

Conclusion

Substitution of water with ethanol in tempura fried squids promotes a more porous, dehydrated and glassy crust structure, which in turn makes the coating crispier and crunchier. Remarkably, these effects on texture remain after storage and reheating. Besides, ethanol based tempuras absorb more oil during frying and seem to develop stronger browning reactions, both facts determining a more intense flavour. On the other hand, incubation of tempuras with CO₂ has not a marked influence on the physical or structural features of the fried coating, although in those in which no ethanol is added, the higher number of bubbles due to the presence of CO₂ seems to improve the crunchiness of the fried coating.

Overall, the addition of ethanol to tempura is an interesting tool for improving crispness and crunchiness of coated fried products, especially in catering or central kitchens, in which storage and reheating are common procedures.

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Compliance with ethical standards

Conflict of interest Authors declare no conflict of interest.

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